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## Introduction

- Study of the performance of the acidic **Ionic Liquid, 1-butyl-3-methylimidazolium hydrogen sulfate** ([BMIM][HSO<sub>4</sub>]), as a catalyst for the esterification of a mix of fatty acids (mostly oleic acid) to the respective FAME's, using methanol.

## Experimental Section

### Reaction

- Esterification takes place in a round flask where the required amounts of oleic acid, acidic catalyst and methanol are added, with reflux and temperature control, during the specified reaction time.
- The product is cooled and decantates for 24 h, with the objective of separating the two phases: an organic phase (biodiesel) and an aqueous phase.

### Acidity Value

- Determination of the acidity value of the biodiesel samples is carried out through acid base volumetric titration with a KOH solution, using a 1:1 diethyl ether/ethanol mix as solvent.

### Gas Chromatography

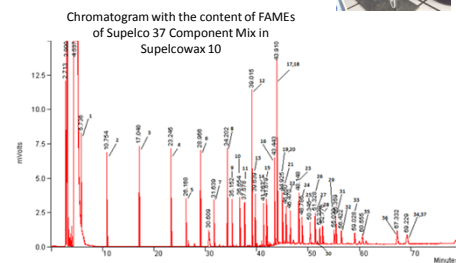
- Analyzing the compounds in a commercial standard mix (37 FAME Mix: 47885-U SUPELCO) serves as a basis for comparison of the retention times of different compounds in biodiesel samples.
- The analysis of the biodiesel samples, using the same operational conditions, allows the qualitative identification of the FAMES present in the sample, and the estimation of the respective content by comparison of areas.



- Multiple reaction batches were designed varying several operational parameters: **temperature, reaction time, oleic acid/methanol mole ratio** and **mass of catalyst**, in order to optimise the reaction yield.
- The prediction of the reaction yield was done using two indirect methods:
  - measuring the final product **acidity value** through volumetric titration,
  - assessing the total **FAME content** of the biodiesel product by Gas Chromatography (GC-FID).



Gas Chromatograph Varian 3800GC equipped with flame ionization detector (FID)



## Results

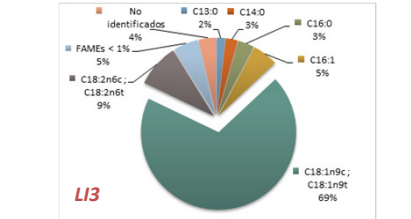
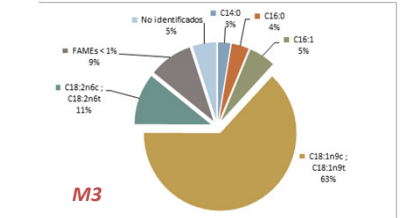
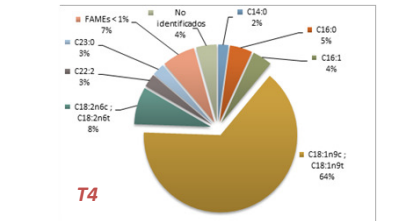
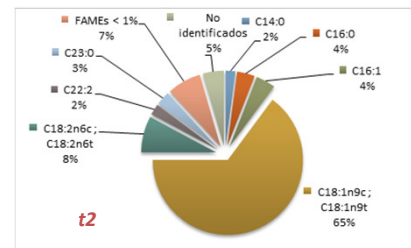
- Optimization of the production of biodiesel

Parameter	Selected value
Reaction time	4 h
Reaction temperature	90 °C
Mole ratio (OA/Met) <sup>a</sup>	1:10
weight % catalyst <sup>b</sup>	10%

<sup>a</sup> OA – oleic acid; Met – methanol  
<sup>b</sup> relating to oleic acid weight

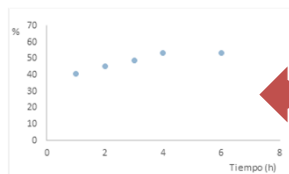
Optimised parameters

- Sample characterisation



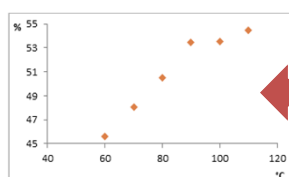
- Reaction yields reach values as high as 90%, and the total weight content of identified FAME's in the biodiesel product is 95-98%.
- The reaction of esterification in the presence of the ionic liquid 1-butyl-3-methyl imidazolium hydrogen sulfate is feasible for the production of biodiesel, but the yields are somewhat lower than those obtained by commonly used catalysts.
- ionic liquid [BMIM]HSO<sub>4</sub> proves to be a promising catalyst for esterification reactions, and a potential alternative for biodiesel production.

### Reaction time



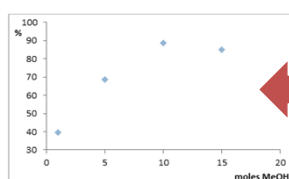
Nº	REACTIVOS	CONDICIONES	ACIDEZ INICIAL (mgKOH/g)	ACIDEZ FINAL (mgKOH/g)	CONVERSION
t1	A.O./MeOH	1 h	194,56	115,16	40,81%
	10% BMIMHSO4	300ppm			
t2	A.O./MeOH	2 h	194,56	106,03	45,51%
	10% BMIMHSO4	300ppm			
t3	A.O./MeOH	3 h	194,56	99,20	49,01%
	10% BMIMHSO4	300ppm			
t4	A.O./MeOH	4 h	194,56	91,01	53,22%
	10% BMIMHSO4	300ppm			
t5	A.O./MeOH	6 h	194,56	90,98	53,24%
	10% BMIMHSO4	300ppm			

### Reaction temperature



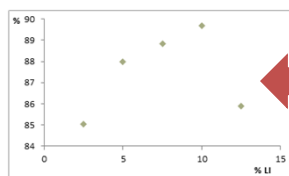
Nº	REACTIVOS	CONDICIONES	ACIDEZ INICIAL (mgKOH/g)	ACIDEZ FINAL (mgKOH/g)	CONVERSION
T1	A.O./MeOH	60°C	184,29	105,80	45,63%
	10% BMIMHSO4	300ppm			
T2	A.O./MeOH	70°C	184,29	101,00	48,07%
	10% BMIMHSO4	300ppm			
T3	A.O./MeOH	80°C	194,56	96,30	50,49%
	10% BMIMHSO4	300ppm			
T4	A.O./MeOH	90°C	194,56	90,60	53,44%
	10% BMIMHSO4	300ppm			
T5	A.O./MeOH	100°C	184,29	90,40	53,52%
	10% BMIMHSO4	300ppm			

### Mole Ratio Oleic Acid/Methanol



Nº	REACTIVOS	CONDICIONES	ACIDEZ INICIAL (mgKOH/g)	ACIDEZ FINAL (mgKOH/g)	CONVERSION
M1	A.O./MeOH	4 h	194,56	117,10	39,79%
	10% BMIMHSO4	300ppm			
M2	A.O./MeOH	4 h	194,56	61,00	68,64%
	10% BMIMHSO4	300ppm			
M3	A.O./MeOH	4 h	194,56	20,10	88,69%
	10% BMIMHSO4	300ppm			
M4	A.O./MeOH	4 h	194,56	29,50	84,82%
	10% BMIMHSO4	300ppm			

### Catalyst %w



Nº	REACTIVOS	CONDICIONES	ACIDEZ INICIAL (mgKOH/g)	ACIDEZ FINAL (mgKOH/g)	CONVERSION
U1	A.O./MeOH	4 h	194,56	29,10	85,03%
	2,5% BMIMHSO4	300ppm			
U2	A.O./MeOH	4 h	194,56	21,40	88,00%
	5% BMIMHSO4	300ppm			
U3	A.O./MeOH	4 h	194,56	21,70	88,84%
	7,5% BMIMHSO4	300ppm			
U4	A.O./MeOH	4 h	194,56	20,10	89,69%
	10% BMIMHSO4	300ppm			
U5	A.O./MeOH	4 h	194,56	27,50	85,88%
	12,5% BMIMHSO4	300ppm			