

Antioxidants in Wild Mushrooms

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Abstract: Maintenance of equilibrium between free radical production and antioxidant defences (enzymatic and non-enzymatic) is an essential condition for normal organism functioning. When this equilibrium has a tendency for the production of free radicals we say that the organism is in oxidative stress. In this situation, excess free radicals may damage cellular lipids, proteins and DNA, affecting normal function and leading to various diseases. In aerobic organisms, the free radicals are constantly produced during the normal cellular metabolism, mainly in the form of Reactive Oxygen Species (ROS) and Reactive Nitrogen Species (RNS). Exposition of the organism to free radicals has led to the development of endogenous defence mechanisms to eliminate them. These defences were the response of evolution to the inevitability of ROS production in aerobic conditions. Natural products with antioxidant activity may help the endogenous defence system. In this perspective the antioxidants present in the diet assume a major importance as possible protector agents reducing oxidative damage. Particularly, the antioxidant properties of wild mushrooms have been extensively studied by our research group and by others, and many antioxidant compounds extracted from these sources have been identified, such as phenolic compounds, tocopherols, ascorbic acid, and carotenoids. We will review the compounds identified so far in mushrooms, as well as the mechanism of action involved in their antioxidant properties. Wild mushrooms might be used directly in diet and promote health, taking advantage of the additive and synergistic effects of all the bioactive compounds present.

OXIDATIVE STRESS

Free radicals are produced in the normal natural metabolism of aerobic cells, mostly in the form of oxygen reactive species (ROS). Once produced, most of the free radicals are neutralized by cellular antioxidant defences (enzymes and non-enzymatic molecules). Maintenance of equilibrium between free radicals production and antioxidant defences is an essential condition for normal organism functioning [1].

Beneficial effects of ROS occur at low or moderate concentrations and involve cellular physiological roles of signalization and regulation [2, 3]. Nevertheless, the equilibrium between ROS production and antioxidant defences might be displaced either by the overproduction of ROS or by the loss of the cell antioxidant defences [4]. This disequilibrium is known as oxidative stress, and in this case, the excess ROS may oxidize and damage cellular lipids, proteins and DNA, leading to their modification and inhibiting their normal function [1, 5, 6].

Oxidative stress might have natural causes such as extreme exercise or inflammation processes, or non-natural causes such as the presence of xenobiotics in the organism or situations related to several diseases (Fig. 1). In fact, the non-controlled production of free radicals has been related to more than one hundred diseases including several kinds of cancer [7], diabetes [1], cirrhoses [8], cardiovascular diseases [9], neurological disorders [10], among others [1]. The overproduction of ROS has also been related to the aging process [11-13].

Considering that 70% of the chronic diseases and related costs can be prevented, the knowledge about ROS and about their overproduction control is crucial [14]. This control can be achieved by the maintenance of good levels of antioxidants and free radicals scavengers, increasing diet quality (higher consumption of vegetables, leguminous and

fruits) or avoiding behaviours that lead to a higher ROS production, such as tobacco, excessive exposure to environmental pollutants and xenobiotics [15].

FREE RADICALS, REACTIVE OXYGEN SPECIES (ROS) AND REACTIVE NITROGEN SPECIES (RNS)

A free radical is defined as any atom or molecule possessing unpaired electrons in the outer orbit [16, 17]. They are generally unstable and very reactive. Free radicals derived from molecular Oxygen (O_2) are usually known by reactive Oxygen species (ROS) and represent the most important class of radical species generated in living systems [18]. In fact, despite the importance of O_2 to aerobic life, in some conditions it can be toxic. This phenomenon is called “oxygen paradox” [19]. Several ROS production pathways and the main endogenous antioxidant defences of the cell are described in Fig. (2).

The addition of one electron to molecular Oxygen forms the superoxide anion ($O_2^{\cdot-}$), which is considered the “primary” ROS. Superoxide anion is mostly produced in mitochondria, due to a small but continuous “leak” of the electrons in the mitochondrial electron transport system (ETS). These electrons generate superoxide anion instead of reducing oxygen to water. Measurements on submitochondrial particles suggest an upper limit of 1–3% of “leaking” electrons in the mitochondrial ETS [20]. Superoxide anion can also be produced by different endogenous enzymatic systems present in the cell like NADPH oxidases and xanthine oxidase [3, 19]. $O_2^{\cdot-}$ have been implicated in several diseases [21].

Even though $O_2^{\cdot-}$ is not a very active radical, it can interact with other molecules generating what are considered as “secondary” ROS, such as hydrogen peroxide (H_2O_2) and hydroxyl radical (OH^{\cdot}). Hydroxyl radical has a very short life time but is considered to be the most toxic among all ROS, being responsible for the attack to DNA molecules, damaging purins and pyrimidines and the structure of desoxyribose DNA [22]. Hydroxyl radical is the neutral form

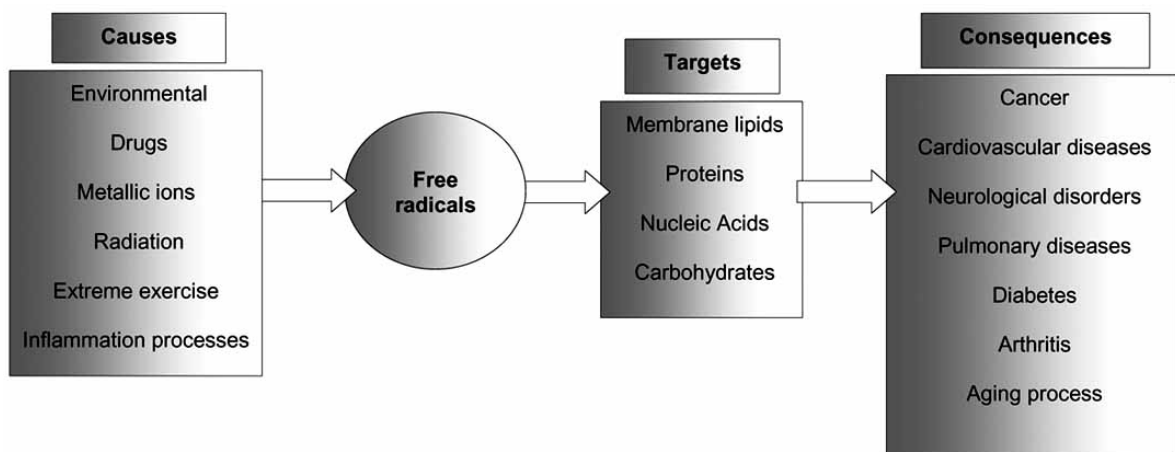


Fig. (1). Major causes for over production of free radical (oxidative stress), possible cellular targets and conditions that were associated to oxidative stress.

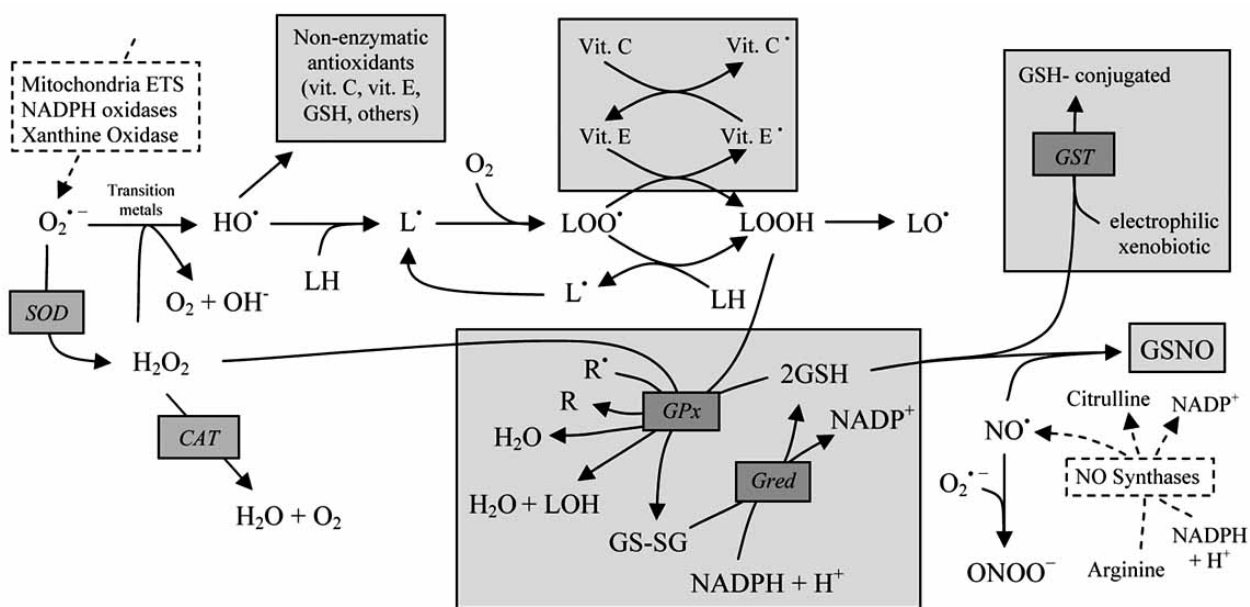


Fig. (2). Overview of the main reactions involving reactive Oxygen species (ROS) / reactive Nitrogen species (RNS), and major endogenous enzymatic and non-enzymatic antioxidant defences in the cell. The most representative endogenous sources (traced rectangles) of ROS/RNS are presented and include: Mitochondrial ETS (Electron transport system), NADPH oxidases, Xanthine oxidase for ROS and NO synthases for RNS. The main antioxidant defences are presented in shaded rectangles and the enzymes involved are presented in *italic*. Molecular Oxygen (O_2), superoxide anion ($O_2^{\bullet-}$), hydrogen peroxide (H_2O_2), hydroxyl radical (HO^\bullet), hydroxide ion (HO^-) membrane lipids (LH), lipid radical (L^\bullet), peroxy radical (LOO^\bullet), hydroperoxide lipid (LOOH), lipid alkoxy radical (LO^\bullet), nitric oxide (NO^\bullet), radicals (R^\bullet), non-radicals (R), alcohols (LOH), glutathione (GSH), glutathione disulphide (GS-SG), α -tocopherol or vitamin E (vit. E), vitamin E radical (vit. E $^\bullet$), vitamin C (vit. C), vitamin C radical (vit. C $^\bullet$), S-nitrosoglutathione (GSNO), nicotinamide adenine dinucleotide phosphate: oxidized (NADP $^+$), reduced (NADPH). Enzymes: Superoxide dismutase (SOD), catalase (CAT), glutathione peroxidase (GPx), glutathione reductase (Gred), glutathione-S-transferases (GST), Mitochondrial ETS (electron transport system), nitric oxide synthase (NOS).

of the hydroxide ion and it is formed by an electron transfer from transition metals to H_2O_2 , and interacts with biomolecules immediately after generation. Eventual permanent damages in the genetic material caused by oxidative stress might represent the first step to mutagenesis, carcinogenesis and aging [1].

Mitochondria are the most important source of ROS, but they are also the first targets of these radicals because ROS

have an easy access to the membrane lipids, which are susceptible to free radicals attack. This attack is called lipid peroxidation and promotes the production of different types of ROS [23] (Fig. 2). The lipid peroxidation usually begins with the extraction of a hydrogen atom from a polyunsaturated lipid (LH) chain, through the action of reactive species such as HO^\bullet . This generates a highly reactive lipid radical (L^\bullet) that can react with O_2 to form a peroxy radical (LOO^\bullet).

If not neutralized by antioxidant defences, the peroxy radical will react with other adjacent lipids producing hydroperoxide lipids (LOOH) that can easily be decomposed to form new L[•] radicals, initiating a process that is known as chain propagation reactions. This process when not stopped, can lead to much superior damage than the ROS that started the reaction [3, 24, 25].

It is also important to notice the existence of radicals with Nitrogen called reactive Nitrogen species (RNS). The principal RNS is nitric oxide (NO[•]) and it is generated in biological tissues by specific nitric oxide synthases (NOS), which metabolise arginine to citrulline (Fig. 2) [3, 26]. NO[•] is an abundant reactive radical that acts as an important oxidative biological signalling molecule in a large variety of physiological processes, including neurotransmission, blood pressure regulation, defence mechanisms, and regulation of immune response [3, 27]. The over expression of RNS is called nitrosative stress and may lead to nitrosylation of proteins and so affect their normal function [6]. Cells of the immune system produce both the superoxide anion and nitric oxide during the oxidative burst triggered in inflammatory processes. Under these conditions, NO[•] can react with O₂^{•-} to produce significant amounts of peroxynitrite anion (ONOO⁻), which is a potent oxidising agent that can cause DNA fragmentation and lipid oxidation [28, 29].

ENDOGENOUS ANTIOXIDANT DEFENCES

Exposure to free radicals from a variety of sources has led organisms to the development of a series of defence mechanisms (Fig. 2) [20]. These defences were the evolution response to the inevitability of the existence of oxygen radicals in aerobic life conditions, and can be classified into enzymatic and non-enzymatic.

There are many different endogenous enzymatic antioxidant defences in the organism, either in intracellular or extracellular medium. Examples of these defences include superoxide dismutase (SOD), catalase (CAT), glutathione peroxidases (GPx), and glutathione reductase (Gred) among others. The endogenous non-enzymatic antioxidant defences include glutathione (GSH), α -tocopherol (vitamin E), ascorbic acid (vitamin C), lipoic acid, and other antioxidants [1, 2, 30]. SOD converts O₂^{•-} into H₂O₂, which is then detoxified to water either by CAT in the peroxisomes or by GPx in the mitochondria, cytosol or nucleus. GPx are a group of selenoenzymes that require Selenium on their biosynthesis. The intake on food or dietary supplements of Selenium is crucial for antioxidant enzyme defences. This is of special importance as mushrooms have been found to contain Selenium in good quantity especially wild edible mushrooms [31]. Another important enzyme is Gred, which regenerates GSH that is used as a hydrogen donor by GPx [32]. GPx can also transform hydroperoxide lipids into alcohols (LOH).

Glutathione (GSH) is a low molecular weight tripeptide composed of glutamate, cysteine, and glycine being the main intracellular redox buffer. The capacity of GSH to regenerate the most important antioxidant molecules is linked with the redox state of the glutathione disulphide/glutathione (GSSG/GSH) couple [33]. GSH effectively scavenges ROS (HO[•], H₂O₂, LOO[•] and ONOO⁻) either directly or indirectly

as a cofactor of several detoxifying enzymes, e.g. GPx, GST, among others. In the neutralization process of ROS, GSH is oxidized to glutathione disulphide (GS-SG), which can be further reduced to two GSH by the enzyme Gred. GSH is also able to regenerate other antioxidant molecules such as vitamins C and E. GSH can also react with a variety of electrophilic xenobiotics in reactions catalysed by glutathione-S-transferases (GST) generating products with higher solubility and thus easier to eliminate. GSH can also neutralize NO[•], resulting in the formation of S-nitrosoglutathione (GSNO) [1, 2].

Vitamin E is a liposoluble vitamin present in the membranes thus playing an important role in the prevention of lipid peroxidation. Among the eight forms of vitamin E, α -tocopherol is the most active form in humans. ROS (hydroxyl and peroxy radicals, etc.) react with vitamin E, generating a poorly reactive phenolic radical (vit. E[•]). Vitamin C then reacts with vit. E[•] producing vitamin C radical (vit. C[•]) and regenerating vitamin E. Both radicals (vit. E[•] and vit. C[•]) are poorly reactive species because of its unpaired electron [2, 34].

Besides all the mentioned endogenous defences, antioxidant supplements or antioxidant-containing foods may be used to help the organism to reduce oxidative damage or to protect food quality by preventing oxidative deterioration [17]. In recent years, the restriction in the use of synthetic antioxidants, such as BHA (2-*tert*-butyl-4-methoxyphenol) and BHT (2,6-di-*tert*-butyl-4-methylphenol), has caused an increased interest towards natural antioxidant substances [35, 36]. Natural antioxidants are being extensively studied for their capacity to protect organisms and cells from damage brought on by oxidative stress.

A multitude of natural antioxidants have already been isolated from different kinds of plant materials such as oilseeds, cereal crops, vegetables, fruits, leaves, roots, spices, and herbs [37]. Epidemiological studies have consistently shown that a high dietary intake of fruits and vegetables is strongly associated with reduced risk of developing chronic diseases, such as cancer and cardiovascular disease [38-42]. In fact, the implication of oxidative and nitrosative stress in the etiology and progression of several acute and chronic clinical disorders has led to the suggestion that antioxidants can have health benefits as prophylactic agents [39]. This suggests that changes in dietary behaviour, increasing consumption of plant-based foods, which contain significant amounts of bioactive phytochemicals, may provide desirable health benefits, beyond basic nutrition, to reduce the risk of chronic diseases [43].

The number of individual phytochemicals already identified in fruits and vegetables is estimated in >5,000, but a large percentage still remains unknown and need to be identified before we can fully understand the health benefits of phytochemicals in whole foods [38].

WILD MUSHROOMS AS A SOURCE OF ANTIOXIDANTS

Mushrooms have become attractive as functional foods and as a source of physiologically beneficial medicine [44-47]. Some advantages of using mushrooms over plants as

Table 1. Studies on Antioxidant Properties of Wild Mushrooms

Mushroom species	Country	Assays	Ref.
<i>Phellinus rimosus</i>	India	1-3	[48]
<i>Ganoderma lucidum</i> , <i>Ganoderma tsugae</i> , <i>Coriolus versicolor</i>	Taiwan	4-8	[49]
<i>Dictophora indusiata</i> , <i>Grifola frondosa</i> , <i>Hericium erinaceus</i> , <i>Trichloma giganteum</i>	Taiwan	4-8	[50]
<i>Terfezia claveryi</i> , <i>Picoa juniperi</i> , <i>Lepista nuda</i> , <i>Lentinus edodes</i> , <i>Agrocybe cylindracea</i> , <i>Cantharellus lutescens</i> , <i>Hydnum repandum</i>	Spain	9	[51]
<i>Flammulina velutipes</i> (white), <i>Flammulina velutipes</i> (yellow), <i>Lentinula edodes</i> , <i>Pleurotus cystidiosus</i> , <i>Pleurotus ostreatus</i>	Taiwan	4-8	[52]
<i>Lentinus edodes</i> , <i>Volvariella volvacea</i>	China	6, 10, 11	[53]
<i>Grifola frondosa</i>	Korea	12	[54]
<i>Phellinus linteus</i>	Korea	3, 6, 13, 14	[55]
<i>Auricularia auricula</i>	India	3, 7	[56]
<i>Grifola frondosa</i> , <i>Morchella esculenta</i> , <i>Termitomyces albuminosus</i>	Taiwan	5-8, 15	[57]
<i>Ganoderma applanatum</i>	India	3, 7	[58]
<i>Lentinus edodes</i> , <i>Volvariella volvacea</i>	China	3	[59]
<i>Inonotus obliquus</i>	Korea	1, 6, 16	[60]
<i>Agrocybe aegerita</i>	China	3, 6, 17	[61]
<i>Lentinus edodes</i>	Korea	6, 17	[62]
<i>Morchella vulgaris</i> , <i>Morchella esculanta</i>	Turkey	1, 5, 6, 8, 18	[63]
<i>Termitomyces heimii</i> , <i>Helvella crispa</i> , <i>Termitomyces tylerance</i> , <i>Lactarius sanguifluus</i> , <i>Morchella conica</i> , <i>Termitomyces mummiformis</i> , <i>Pleurotus sajor-caju</i> , <i>Termitomyces shimperi</i> , <i>Lentinus squarulosus</i> , <i>Boletus edulis</i> , <i>Pleurotus djamor</i> , <i>Macrolepiota procera</i> , <i>Cantharellus clavatus</i> , <i>Morchella angusticeps</i> , <i>Termitomyces microcarpus</i> , <i>Lactarius deliciosus</i> , <i>Geastrum arinarius</i> , <i>Hydnum repandum</i> , <i>Lentius sajor-caju</i> , <i>Sparassis crispa</i> , <i>Russula brevipes</i> , <i>Auricularia polytricha</i> , <i>Cantharellus cibarius</i>	India	3, 5, 6	[64]
<i>Suillus bellini</i> , <i>Tricholomopsis rutilans</i> , <i>Hygrophorus agathosmus</i> , <i>Amanita rubescens</i> , <i>Russula cyanoxantha</i> , <i>Boletus edulis</i> , <i>Tricholoma equestre</i> , <i>Suillus luteus</i> , <i>Suillus granulatus</i>	Portugal	6	[65]
<i>Pleurotus citrinopileatus</i>	Taiwan	5, 6	[66]
<i>Agaricus arvensis</i> , <i>Leucopaxillus giganteus</i> , <i>Sarcodon imbricatus</i>	Portugal	5, 6, 10, 11	[67]
<i>Lactarius piperatus</i>	Portugal	5, 6, 10, 11	[68]
<i>Geastrum saccatum</i>	Brazil	1, 3, 7	[69]
<i>Agaricus bisporus</i> , <i>Polyporus squamosus</i> , <i>Pleurotus ostreatus</i> , <i>Lepista nuda</i> , <i>Russula delica</i> , <i>Boletus badius</i> , <i>Verpa conica</i>	Turkey	1, 5, 6, 8, 19	[70]
<i>Lactarius deliciosus</i> , <i>Sarcodon imbricatus</i>	Portugal	5, 6	[71]
<i>Lentinula edodes</i>	Brazil	6	[72]
<i>Hypsizigus marmoreus</i>	Taiwan	5-8, 15, 20	[73]
<i>Armillariella mellea</i>	Taiwan	3	[74]
<i>Agaricus blazei</i>	Brazil	6, 17	[75]
<i>Agaricus blazei</i> , <i>Agrocybe cylindracea</i> , <i>Boletus edulis</i>	Taiwan	5-8, 15	[76]
<i>Laetiporus sulphureus</i>	Turkey	6, 10	[77]
<i>Agaricus arvensis</i> , <i>Agaricus bisporus</i> , <i>Agaricus silvicola</i> , <i>Agaricus silvaticus</i> , <i>Agaricus romagnesii</i>	Portugal	3, 5, 6, 10, 11	[78]
<i>Cantharellus cibarius</i> , <i>Hypholoma fasciculare</i> , <i>Lepista nuda</i> , <i>Lycoperdon molle</i> , <i>Lycoperdon perlatum</i> , <i>Ramaria botrytis</i> , <i>Tricholoma acerbum</i>	Portugal	3, 5, 6, 10	[79]
<i>Lactarius deliciosus</i> , <i>Macrolepiota mastoidea</i> , <i>Macrolepiota procera</i> , <i>Sarcodon imbricatus</i>	Portugal	3, 5, 6, 10, 11	[80]
<i>Pleurotus ostreatus</i>	India	1, 3, 5, 7, 8	[81]
<i>Pleurotus ostreatus</i> , <i>Agaricus bisporus</i> , <i>Flammulina velutipes</i> , <i>Pleurotus eryngii</i> , <i>Lentinus edodes</i> , <i>Agaricus blazei</i> , <i>Sparassis crispa</i> , <i>Phellinus linteus</i> , <i>Ganoderma lucidum</i> , <i>Inonotus obliquus</i>	Korea	6	[82]
<i>Lactarius deterrimus</i> , <i>Suillus collitinus</i> , <i>Boletus edulis</i> , <i>Xerocomus chrysenteron</i>	Turkey	5, 6, 8, 10	[83]
<i>Agaricus blazei</i>	Brazil	6, 8, 10	[84]

1- Superoxide anion radical scavenging activity; 2- Nitric oxide scavenging activity; 3- Thiobarbituric reactive substances (TBARS assay); 4- 1,3-diethyl-2-thiobarbituric acid (DETBA) method; 5- Reducing power; 6- 2,2-Diphenyl-1-picrylhydrazyl (DPPH) scavenging activity; 7- Hydroxyl radicals scavenging activity; 8- Chelating effects on ferrous ions; 9- Linoleic acid assay; 10- β -carotene bleaching inhibition; 11- Hemolysis inhibition; 12- SOD activity; 13- Xanthine oxidase inhibition; 14- Chorioallantoic membrane (CAM) assay; 15- Conjugated diene method; 16- 2,7-dichlorofluoresceindiacetate (DCF)/2,2'-azobis(2-amidinopropane)dihydrochloride (AAPH) assay; 17- 2,2'-azinobis-(3-ethylbenzthiazoline-6-sulphonic acid (ABTS^{•+}) radical cation scavenging activity; 18- Hydrogen peroxide scavenging activity; 19- Thiocyanate method; 20- Chelating effects on cupric ions.

sources of bioactive compounds are that often the fruiting body can be produced in much less time, the mycelium may also be rapidly produced in liquid culture and the culture medium can be manipulated to produce optimal quantities of active products.

Different wild mushroom species were reported to have antioxidant activity, which was mainly related to their phenolic content [48-84]. Several protocols were used to determine the mushrooms antioxidant activity (Table 1).

There are some works reporting the influence of processing conditions, growth conditions and growing stages of mushrooms, on the antioxidant potential. Our research group [80] described that the amount of antioxidants in cooked samples of *Lactarius deliciosus*, *Macrolepiota mastoidea*, *Macrolepiota procera* and *Sarcodon imbricatus* significantly decreased. We explained that heat used in the cooking procedure could destroy the structures of polyphenols and cause a decrease in their antioxidant activity. Nevertheless, at low heating temperatures, an increase in phenolics concentration may occur. This was observed by us in the dried mushrooms and by Choi *et al.* [62] who described that heat treatment of *Lentinus edodes* increased the overall content of free polyphenolic and flavonoid compounds. The authors suggested that heat treatment might produce changes in their extractability due to the disruption of the cell wall thus bound polyphenolic and flavonoid compounds may be released more easily relative to those of raw materials. Another reason for the improved antioxidant activity could be the formation of novel compounds having antioxidant activities during heat treatment or thermal processing. Valentão *et al.* [85] described that the conservation procedures such as preservation in olive oil or vinegar, seem to affect the qualitative and quantitative phenolics and organic acids profiles of *Cantharellus cibarius*. Particularly, ascorbic acid increased significantly in the samples preserved in vinegar. The mushrooms growth stage also influences their antioxidant contents [86]. In fact, fruiting bodies of *Lactarius deliciosus* and *Lactarius piperatus* in a mature stage with mature spores, revealed lower content in antioxidants such as phenolics, ascorbic acid and β -carotene. This was explained with the involvement of those compounds in defence mechanisms inherent to the aging process (presence of mature spores), resulting in the lowering of their contents in the most advanced stage.

The antioxidants found in mushrooms are mainly phenolic compounds (phenolic acids and flavonoids), followed by tocopherols, ascorbic acid and carotenoids. These molecules were quantified in tens of different species mainly from Finland, India, Korea, Poland, Portugal, Taiwan and Turkey (Table 2). The values are available in literature, but expressed in different basis (dry weight, fresh weight and extract). *Helvella crispa* from India revealed the highest content of phenolic compounds expressed per g of extract (34.65 mg/g), while *Sparassis crispa* from Korea revealed the highest value expressed in a dry weight basis (0.76 mg/g). *Auricularia fuscusuccinea* (white) from Taiwan (32.46 mg/g of extract), *Agaricus silvaticus* (3.23×10^{-3} mg/g of dry weight) and *Ramaria Botrytis* (2.50×10^{-4} mg/g of fresh weight) from Portugal, were the richest species in tocopherols. *Auricularia fuscusuccinea* (brown) from Taiwan

(11.24 mg/g of extract), *Suillus collinitus* from Portugal (3.79 mg/g of dry weight) and *Agaricus bisporus* from Poland (0.22 mg/g of fresh weight) revealed the highest levels of ascorbic acid. *Lactarius deliciosus* from Portugal revealed the highest contents in β -carotene (0.09 mg/g of extract).

Phenolic Compounds

There are a large number of manuscripts reporting determination of total phenolics by Folin Ciocalteu's assay [48-84]. Phenolic compounds include different subclasses (flavonoids, phenolic acids, stilbenes, lignans, tannins, oxidized polyphenols) displaying a large diversity of structures, some of which may escape the usual methodologies of analysis, commonly carried out by HPLC (High Performance Liquid Chromatography) coupled to distinct detection devices. Various reasons exist for that, like the existence of isomers, difficulty for chromatographic separation of some compounds, lack of commercial standards, or structure not yet elucidated [96]. The method of Folin Ciocalteu's is, therefore, largely used to evaluate total phenolics despite all the interferences of this assay since the reagent (mixture of phosphotungstic acid and phosphomolibdic acid) also reacts with other non-phenolic reducing compounds leading to an overvaluation of the phenolic content. For instance, ascorbic acid is a widespread reducing agent that can interfere in the Folin-Ciocalteu's reaction [96]. Other reducing substances such as some sugars and amino acids could also interfere. In addition, the results have to be expressed in equivalents of a particular standard compound (like catechin, gallic acid or tannin acid). All these aspects increase the importance of the determination of phenolic compounds by more sophisticated techniques.

A few studies concerning the analysis of the phenolic components of wild mushrooms can be found in the literature [64, 65, 81, 82, 85, 88, 89, 92]. The individual profiles of phenolic compounds were obtained by high-performance liquid chromatography coupled to photodiode array detector (HPLC-DAD) [64, 65, 82, 85, 88, 92], or to an ultraviolet detector [81], or by gas chromatography-mass spectrometry selected ion monitoring (GC-MS SIM) [89]. Table 3 presents the phenolic compounds found in several mushroom species.

Phenolic compounds are aromatic hydroxylated compounds, possessing one or more aromatic rings with one or more hydroxyl groups, being commonly found in vegetables, fruits and many food sources that form a significant portion of our diet, and some of which are among the most potent and therapeutically useful bioactive substances [97]. Natural phenolic compounds accumulate as end-products from the shikimate and acetate pathways and can range from relatively simple molecules (phenolic acids, phenylpropanoids, flavonoids) to highly polymerised compounds (lignins, melanins, tannins), with flavonoids representing the most common and widely distributed sub-group [98].

As described above, the main phenolic compounds found in mushrooms were phenolic acids. Phenolic acids can be divided into two major groups, hydroxybenzoic acids (Fig. 3) and hydroxycinnamic acids (Fig. 4), which are derived from non-phenolic molecules benzoic and cinnamic acid, respectively.

Table 2. Antioxidants Quantified in Wild Mushrooms

Mushroom species	Phenolic compounds	Tocopherols	Ascorbic acid	β-Carotene	Country	Ref.
<i>Agaricus arvensis</i>	0.17 ^a	1.22 × 10 ^{-3 a}	0.02 ^c	8.52 × 10 ^{-3c}	Portugal	[78,87,88]
<i>Agaricus bisporus (white)</i>	4.32 × 10 ^{-3 a}	-	0.17 ^a	-	Finland	[89]
<i>Agaricus bisporus (brown)</i>	4.69 × 10 ^{-3 a}	-	0.21 ^a	-	Finland	[89]
<i>Agaricus bisporus</i>	0.54 ^a	-	-	-	Korea	[82]
<i>Agaricus bisporus</i>	-	-	0.22 ^b	-	Poland	[90]
<i>Agaricus bisporus</i>	0.03 ^a	2.41 × 10 ^{-3 a}	0.03 ^c	1.95 × 10 ^{-3 c}	Portugal	[78,87,88]
<i>Agaricus bisporus</i>	-	9.20 ^c	-	0.04 ^c	Turkey	[70]
<i>Agaricus blazei</i>	0.70 ^a	-	-	-	Korea	[82]
<i>Agaricus blazei</i>	-	5.44 ^c	-	-	Taiwan	[76]
<i>Agaricus romagnesii</i>	0.08 ^a	1.29 × 10 ^{-3 a}	0.04 ^c	1.32 × 10 ^{-3c}	Portugal	[78,88]
<i>Agaricus silvaticus</i>	-	3.23 × 10 ^{-3 a}	0.04 ^c	5.42 × 10 ^{-3c}	Portugal	[78,87]
<i>Agaricus silvicola</i>	0.35 ^a	1.17 × 10 ^{-3 a}	0.04 ^c	3.02 × 10 ^{-3c}	Portugal	[78,87,88]
<i>Agrocybe cylindracea</i>	-	5.27 ^c	-	-	Taiwan	[76]
<i>Amanita caesarea</i>	-	-	2.07 ^a	-	Portugal	[91]
<i>Amanita rubescens</i>	0.49 ^a	-	0.03 ^a	-	Portugal	[65,92]
<i>Auricularia mesenterica</i>	-	9.45 ^c	1.63 ^c	-	Taiwan	[93]
<i>Auricularia fuscusuccinea (brown)</i>	-	12.69 ^c	11.24 ^c	-	Taiwan	[93]
<i>Auricularia fuscusuccinea (white)</i>	-	32.46 ^c	7.99 ^c	-	Taiwan	[93]
<i>Auricularia polytricha</i>	3.17 ^c	-	-	-	India	[64]
<i>Auricularia polytricha</i>	-	23.61 ^c	3.28 ^c	-	Taiwan	[93]
<i>Boletus badius</i>	-	8.80 ^c	-	-	Turkey	[70]
<i>Boletus edulis</i>	10.19 ^c	-	-	-	India	[64]
<i>Boletus edulis</i>	-	3.30 × 10 ^{-4a}	-	2.73 × 10 ^{-3c}	Portugal	[94]
<i>Boletus edulis</i>	-	6.18 ^c	-	-	Taiwan	[76]
<i>Calocybe gambosa</i>	-	4.00 × 10 ^{-4a}	0.40 ^c	6.41 × 10 ^{-3c}	Portugal	[94]
<i>Calvatia gigantea</i>	-	-	0.15 ^a	-	India	[95]
<i>Cantharellus cibarius</i>	2.00 ^c	3.00 × 10 ^{-5a}	0.42 ^a	-	India	[64,95]
<i>Cantharellus cibarius</i>	7.80 × 10 ⁻³ to 2.54 × 10 ^{-2a}	1.50 × 10 ^{-4a}	0.48 ^c	0.01 ^c	Portugal	[50,85,88]
<i>Cantherallus clavatus</i>	13.22 ^c	-	-	-	India	[64]
<i>Clavulina cinerea</i>	-	-	0.42 ^a	-	India	[95]
<i>Craterellus cornucopioides</i>	-	1.87 × 10 ^{-3 a}	0.87 ^c	0.01 ^c	Portugal	[94]
<i>Fistulina hepatica</i>	0.37 to 0.55 ^a	-	2.80 ^a	-	Portugal	[92]
<i>Flammulina velutipes</i>	0.17 ^a	-	-	-	Korea	[82]
<i>Ganoderma lucidum</i>	0.16 ^a	-	-	-	Korea	[82]
<i>Ganoderma lucidum</i>	-	1.19 ^c	-	-	Taiwan	[49]
<i>Ganoderma tsugae</i>	-	1.07 ^c	-	-	Taiwan	[49]
<i>Gastrum arinarius</i>	4.80 ^c	-	-	-	India	[64]
<i>Gomphus floccosus</i>	-	-	0.26 ^a	-	India	[91]
<i>Grifola frondosa</i>	-	0.05, 0.11 ^c	0.05, 0.14 ^c	-	Taiwan	[50,57]
<i>Helvella crispa</i>	34.65 ^c	-	-	-	India	[64]
<i>Hericium erinaceus</i>	-	0.06 ^c	-	-	Taiwan	[50]

(Table 2). Contd.....

Mushroom species	Phenolic compounds	Tocopherols	Ascorbic acid	β -Carotene	Country	Ref.
<i>Hydnum repandum</i>	7.40 ^c	-	-	-	India	[64]
<i>Hypholoma fasciculare</i>	-	6.00×10^{-5b}	0.09 ^c	0.02 ^c	Portugal	[79]
<i>Hypsizigus marmoreus</i>	-	2.96 ^c	0.13 ^c	0.02 ^c	Taiwan	[74]
<i>Ionotus obliquus</i>	0.55 ^a	-	-	-	Korea	[82]
<i>Lactarius deliciosus</i>	7.32 ^c	-	-	-	India	[64]
<i>Lactarius deliciosus</i>	0.02 ^a	-	0.19 to 0.97 ^a	0.09 ^c	Portugal	[86,88,91]
<i>Lactarius piperatus</i>	-	-	0.16 ^c	0.03 ^c	Portugal	[68]
<i>Lactarius quieticolor</i>	-	-	0.18 ^a	-	India	[95]
<i>Lactarius sangifluus</i>	14.90 ^c	-	-	-	India	[64]
<i>Lentinula edodes</i>	-	0.13 ^c	-	-	Taiwan	[52]
<i>Lentinus edodes</i>	0.01 ^a	-	0.25 ^a	-	Finland	[89]
<i>Lentinus edodes</i>	0.03 ^a	-	-	-	Korea	[82]
<i>Lentinus sajor caju</i>	6.44 ^c	-	-	-	India	[64]
<i>Lentinus squarulosus</i>	15.00 ^c	-	-	-	India	[64]
<i>Lepista nuda</i>	0.07 ^a	9.00×10^{-5b}	0.23 ^c	2.52×10^{-3c}	Portugal	[79,88]
<i>Lepista nuda</i>	-	1.40 ^c	-	7.00×10^{-3c}	Turkey	[70]
<i>Leucopaxillus giganteus</i>	-	-	0.13 ^c	1.88×10^{-3c}	Portugal	[67]
<i>Lycoperdon molle</i>	0.08 ^a	3.00×10^{-5b}	0.34 ^c	4.48×10^{-3c}	Portugal	[79,88]
<i>Lycoperdon perlatum</i>	0.01 ^a	3.00×10^{-5b}	0.21 ^c	0.01 ^c	Portugal	[79,88]
<i>Macrolepiota procera</i>	10.00 ^c	-	-	-	India	[64]
<i>Macrolepiota procera</i>	0.02 ^a	-	-	-	Portugal	[88]
<i>Marasmius oreades</i>	-	2.50×10^{-4a}	-	1.99×10^{-3c}	Portugal	[94]
<i>Morchella anguiticeps</i>	12.92 ^c	-	-	-	India	[64]
<i>Morchella conica</i>	16.90 ^c	-	-	-	India	[64]
<i>Morchella esculenta</i>	-	0.07 ^c	0.13 ^c	-	Taiwan	[57]
<i>Phellinus linteus</i>	0.21 ^a	-	-	-	Korea	[82]
<i>Pleurotus eryngii</i>	0.03 ^a	-	-	-	Korea	[82]
<i>Pleurotus cystidiosus</i>	-	0.45 ^c	-	-	Taiwan	[52]
<i>Pleurotus djamor</i>	13.22 ^c	-	-	-	India	[64]
<i>Pleurotus ostreatus</i>	-	-	0.20 ^a	-	Finland	[89]
<i>Pleurotus ostreatus</i>	0.71 ^a	0.30 ^c	0.25 ^c	0.03 ^c	India	[81]
<i>Pleurotus ostreatus</i>	0.09 ^a	-	-	-	Korea	[82]
<i>Pleurotus ostreatus</i>	-	0.24 ^c	-	-	Taiwan	[52]
<i>Pleurotus ostreatus</i>	-	0.90 ^c	-	-	Turkey	[70]
<i>Pleurotus sajor-caju</i>	14.43 ^c	-	-	-	India	[64]
<i>Polyporus squamosus</i>	-	0.30 ^c	-	0.02 ^c	Turkey	[70]
<i>Ramaria botrytis</i>	0.36 ^a	2.50×10^{-4b}	0.27 ^c	0.01 ^c	Portugal	[79,88]
<i>Ramaria brevispora</i>	-	-	0.28 ^a	-	India	[95]
<i>Russula brevipes</i>	5.50 ^c	-	-	-	India	[64]
<i>Russula delica</i>	-	4.20 ^c	-	9.00×10^{-3c}	Turkey	[70]
<i>Russula integra</i>	-	-	0.20 ^a	-	India	[95]
<i>Sarcodon imbricatus</i>	0.03 ^a	-	0.16 ^c	2.53×10^{-3c}	Portugal	[67,88]

(Table 2). Contd.....

Mushroom species	Phenolic compounds	Tocopherols	Ascorbic acid	β-Carotene	Country	Ref.
<i>Sparassis crispa</i>	5.50 ^c	-	-	-	India	[64]
<i>Sparassis crispa</i>	0.76 ^a	-	-	-	Korea	[82]
<i>Suillus granulatus</i>	2.00 × 10 ⁻³ to 1.59 × 10 ^{-2a}	-	-	-	Portugal	[65,92]
<i>Suillus collinitus</i>	-	-	0.92 to 3.79 ^a	-	Portugal	[91]
<i>Suillus luteus</i>	4.6 × 10 ^{-3a}	-	-	-	Portugal	[65]
<i>Termitomyces albuminosus</i>	-	0.10 ^c	0.13 ^c	-	Taiwan	[57]
<i>Termitomyces heimii</i>	37.00 ^c	-	-	-	India	[64]
<i>Termitomyces microcarpus</i>	6.73 ^c	-	-	-	India	[64]
<i>Termitomyces mummiformis</i>	19.20 ^c	-	-	-	India	[64]
<i>Termitomyces shimperi</i>	15.20 ^c	-	-	-	India	[64]
<i>Termitomyces tylerance</i>	17.88 ^c	-	-	-	India	[64]
<i>Tremella fuciformis</i>	-	22.10 ^c	6.74 ^c	-	Taiwan	[93]
<i>Tricholoma acerbum</i>	0.04 ^a	8.00 × 10 ^{-5b}	0.22 ^c	0.08 ^c	Portugal	[79,88]
<i>Tricholoma equestre</i>	0.04, 0.12 ^a	-	-	-	Portugal	[65]
<i>Tricholoma giganteum</i>	-	0.12 ^c	-	-	Taiwan	[50]
<i>Verpa conica</i>	-	1.90 ^c	-	0.01 ^c	Turkey	[70]

Values are Presented in: ^amg/g of Dry Weight; ^bmg/g of Fresh Weight; ^cmg/g of Extract.

Table 3. Phenolic Compounds Detected in Wild Mushrooms

Phenolic compound	Mushroom species	Country	Ref.
Benzoic acid	<i>Agaricus blazei</i> , <i>Sparassis crispa</i> , <i>Phellinus linteus</i>	Korea	[82]
<i>p</i> -Hydroxybenzoic acid	<i>Agaricus bisporus</i> (white), <i>Agaricus bisporus</i> (brown), <i>Lentinus edodes</i>	Finland	[89]
	<i>Amanita rubescens</i> , <i>Russula cyanoxantha</i> , <i>Tricholoma equestre</i>	Portugal	[65]
	<i>Amanita rubescens</i> , <i>Suillus granulatus</i>	Portugal	[92]
	<i>Agaricus arvensis</i> , <i>Agaricus bisporus</i> , <i>Agaricus silvicola</i> , <i>Agaricus romagnesii</i> , <i>Lactarius deliciosus</i> , <i>Lepista nuda</i> , <i>Lycoperdon molle</i> , <i>Sarcodon imbricatus</i> , <i>Ramaria botrytis</i> , <i>Tricholoma acerbum</i>	Portugal	[88]
	<i>Sparassis crispa</i> , <i>Phellinus linteus</i> , <i>Ionotus obliquus</i>	Korea	[82]
Protocatechuic acid	<i>Agaricus bisporus</i> (white), <i>Agaricus bisporus</i> (brown), <i>Lentinus edodes</i>	Finland	[89]
	<i>Termitomyces heimii</i> , <i>Termitomyces mummiformis</i> , <i>Boletus edulis</i> , <i>Lactarius deliciosus</i> , <i>Pleurotus sajor-caju</i> , <i>Hydnum repandum</i> , <i>Lentinus squarrosulus</i> , <i>Sparassis crispa</i> , <i>Morchella conica</i> , <i>Russula brevipes</i> , <i>Lactarius sangifluus</i> , <i>Macrolepiota procera</i> , <i>Cantherallus clavatus</i> , <i>Auricularia polytricha</i> , <i>Pleurotus djamor</i> , <i>Lentinus sajor caju</i> , <i>Termitomyces tylerance</i> , <i>Morchella anguiticeps</i> , <i>Termitomyces microcarpus</i> , <i>Helvella crispa</i> , <i>Termitomyces shimperi</i>	India	[64]
	<i>Lepista nuda</i> , <i>Ramaria botrytis</i>	Portugal	[88]
	<i>Pleurotus ostreatus</i> , <i>Agaricus bisporus</i> , <i>Flammulina velutipes</i> , <i>Pleurotus eryngii</i> , <i>Lentinus edodes</i> , <i>Agaricus blazei</i> , <i>Sparassis crispa</i> , <i>Phellinus linteus</i> , <i>Ganoderma lucidum</i> , <i>Ionotus obliquus</i>	Korea	[82]
Gallic acid	<i>Termitomyces heimii</i> , <i>Termitomyces mummiformis</i> , <i>Lactarius deliciosus</i> , <i>Pleurotus sajor-caju</i> , <i>Hydnum repandum</i> , <i>Lentinus squarrosulus</i> , <i>Sparassis crispa</i> , <i>Morchella conica</i> , <i>Russula brevipes</i> , <i>Geastrum arinarius</i> , <i>Cantharellus cibarius</i> , <i>Lactarius sangifluus</i> , <i>Macrolepiota procera</i> , <i>Cantherallus clavatus</i> , <i>Auricularia polytricha</i> , <i>Pleurotus djamor</i> , <i>Lentinus sajor caju</i> , <i>Termitomyces tylerance</i> , <i>Morchella anguiticeps</i> , <i>Termitomyces Microcarpus</i> , <i>Helvella crispa</i> , <i>Termitomyces shimperi</i>	India	[64]
	<i>Pleurotus ostreatus</i> , <i>Agaricus bisporus</i> , <i>Flammulina velutipes</i> , <i>Pleurotus eryngii</i> , <i>Lentinus edodes</i> , <i>Agaricus blazei</i> , <i>Sparassis crispa</i> , <i>Phellinus linteus</i> , <i>Ganoderma lucidum</i> , <i>Ionotus obliquus</i>	Korea	[82]
Gentisic acid	<i>Termitomyces heimii</i> , <i>Termitomyces mummiformis</i> , <i>Lactarius deliciosus</i> , <i>Pleurotus sajor-caju</i> , <i>Hydnum repandum</i> , <i>Lentinus squarrosulus</i> , <i>Sparassis crispa</i> , <i>Morchella conica</i> , <i>Russula brevipes</i> , <i>Lactarius sangifluus</i> , <i>Macrolepiota procera</i> , <i>Cantherallus clavatus</i> , <i>Auricularia polytricha</i> , <i>Pleurotus djamor</i> , <i>Termitomyces tylerance</i> , <i>Morchella anguiticeps</i> , <i>Termitomyces Microcarpus</i> , <i>Helvella crispa</i> , <i>Termitomyces shimperi</i>	India	[64]
	<i>Agaricus blazei</i>	Korea	[82]

(Table 3). Contd.....

Phenolic compound	Mushroom species	Country	Ref.
Homogentisic acid	<i>Pleurotus ostreatus, Flammulina velutipes, Ionotus obliquus</i>	Korea	[82]
Vanillic acid	<i>Termitomyces heimii, Pleurotus sajorcaju, Hydnum repandum, Lentinus squarulosus, Morchella conica, Russula brevipes, Lactarius sangifluus, Macrolepiota procera, Cantharellus clavatus, Auricularia polytricha, Pleurotus djamor, Lentinus sajor caju, Termitomyces Microcarpus, Helvella crispa, Termitomyces shimperi</i>	India	[64]
	<i>Lycoperdon molle, Tricholoma acerbum</i>	Portugal	[88]
5-Sulfosalicylic acid	<i>Flammulina velutipes, Sparassis crispa, Phellinus linteus, Ganoderma lucidum</i>	Korea	[82]
Syringic acid	<i>Termitomyces mummiformis, Hydnum repandum, Morchella conica, Russula brevipes, Lactarius sangifluus, Macrolepiota procera, Cantharellus clavatus, Pleurotus djamor, Lentinus sajor caju, Termitomyces tylerance, Morchella anguiticeps, Termitomyces Microcarpus</i>	India	[64]
	<i>Agaricus blazei, Sparassis crispa</i>	Korea	[82]
Veratric acid	<i>Sparassis crispa</i>	Korea	[82]
Vanillin	<i>Ionotus obliquus</i>	Korea	[82]
Cinnamic acid	<i>Agaricus bisporus (white), Agaricus bisporus (brown), Lentinus edodes</i>	Finland	[89]
	<i>Termitomyces heimii, Termitomyces mummiformis, Pleurotus sajor-caju, Hydnum repandum, Lentinus squarulosus, Sparassis crispa, Lactarius sangifluus, Cantharellus clavatus, Pleurotus djamor, Termitomyces shimperi</i>	India	[64]
	<i>Agaricus arvensis, Agaricus bisporus, Agaricus silvicola, Agaricus romagnesii, Cantharellus cibarius, Lycoperdon perlatum, Macrolepiota procera</i>	Portugal	[88]
	<i>Agaricus blazei</i>	Korea	[82]
<i>p</i> -Coumaric acid	<i>Cantharellus cibarius</i>	Portugal	[85]
	<i>Termitomyces heimii, Boletus edulis, Sparassis crispa, Geastrum arinarius, Cantharellus cibarius, Lactarius sangifluus, Macrolepiota procera, Pleurotus djamor, Lentinus sajor caju</i>	India	[64]
	<i>Fistulina hepatica</i>	Portugal	[92]
	<i>Agaricus arvensis, Agaricus silvicola, Lepista nuda</i>	Portugal	[88]
	<i>Sparassis crispa</i>	Korea	[82]
<i>o</i> -Coumaric acid	<i>Ionotus obliquus</i>	Korea	[82]
Caffeic acid	<i>Sparassis crispa</i>	Korea	[82]
	<i>Cantharellus cibarius</i>	Portugal	[85]
	<i>Termitomyces heimii, Boletus edulis, Lentinus squarulosus, Morchella conica, Russula brevipes, Cantharellus cibarius, Lactarius sangifluus, Macrolepiota procera, Cantharellus clavatus, Pleurotus djamor, Lentinus sajor caju, Termitomyces tylerance, Morchella anguiticeps, Termitomyces Microcarpus, Termitomyces shimperi</i>	India	[64]
	<i>Fistulina hepatica</i>	Portugal	[92]
	<i>Flammulina velutipes, Sparassis crispa, Phellinus linteus</i>	Korea	[82]
Ferulic acid	<i>Termitomyces heimii, Lactarius deliciosus, Pleurotus sajor-caju, Lentinus squarulosus, Sparassis crispa, Morchella conica, Cantharellus cibarius, Lactarius sangifluus, Macrolepiota procera, Cantharellus clavatus, Pleurotus djamor, Termitomyces Microcarpus, Termitomyces shimperi</i>	India	[64]
	<i>Flammulina velutipes, Ionotus obliquus</i>	Korea	[82]
3- <i>O</i> -Caffeoylquinic acid	<i>Cantharellus cibarius</i>	Portugal	[85]
4- <i>O</i> -Caffeoylquinic acid	<i>Cantharellus cibarius</i>	Portugal	[85]
5- <i>O</i> -Caffeoylquinic acid	<i>Cantharellus cibarius</i>	Portugal	[85]
	<i>Pleurotus ostreatus, Flammulina velutipes, Phellinus linteus</i>	Korea	[82]
Quercetin	<i>Suillus luteus, Suillus granulatus</i>	Portugal	[65]
	<i>Flammulina velutipes, Agaricus blazei, Sparassis crispa, Ganoderma lucidum, Ionotus obliquus</i>	Korea	[82]
Rutin	<i>Cantharellus cibarius</i>	Portugal	[85]
	<i>Pleurotus ostreatus</i>	India	[81]
Kaempferol	<i>Sparassis crispa, Ganoderma lucidum, Ionotus obliquus</i>	Korea	[82]

(Table 3). Contd.....

Phenolic compound	Mushroom species	Country	Ref.
Myricetin	<i>Pleurotus ostreatus, Agaricus bisporus, Agaricus blazei, Ganoderma lucidum</i>	Korea	[82]
Chrysin	<i>Pleurotus ostreatus</i>	India	[81]
Catechin	<i>Lentinus edodes, Agaricus blazei, Ganoderma lucidum</i>	Korea	[82]
Hesperetin	<i>Ganoderma lucidum</i>	Korea	[82]
Naringenin	<i>Sparassis crispa</i>	Korea	[82]
Naringin	<i>Pleurotus ostreatus, Agaricus bisporus, Pleurotus eryngii, Ganoderma lucidum, Ionotus obliquus</i>	Korea	[82]
Formometin	<i>Ganoderma lucidum</i>	Korea	[82]
Biochanin	<i>Ganoderma lucidum</i>	Korea	[82]
Pyrogallol	<i>Agaricus bisporus, Flammulina velutipes, Agaricus blazei, Sparassis crispa, Ganoderma lucidum, Phellinus linteus</i>	Korea	[82]
Resveratrol	<i>Sparassis crispa, Inonotus obliquus</i>	Korea	[82]
Ellagic acid	<i>Fistulina hepatica</i>	Portugal	[92]
Tannic acid	<i>Termitomyces heimii, Termitomyces mummiformis, Boletus edulis, Lactarius deliciosus, Pleurotus sajor-caju, Hydnum repandum, Lentinus squarulosus, Morchella conica, Russula brevipes, Geastrum arinarius, Cantharellus cibarius, Lactarius sangifluus, Macrolepiota procera, Cantharellus clavatus, Auricularia polytricha, Pleurotus djamor, Termitomyces tylerance, Morchella anguiticeps, Termitomyces Microcarpus, Helvella crispa, Termitomyces shimperi</i>	India	[64]

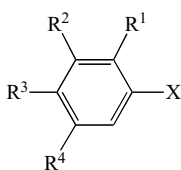
Benzoic acid derivatives	Substitution					
	X	R ¹	R ²	R ³	R ⁴	
<i>p</i> -Hydroxybenzoic	COOH	H	H	H	OH	
Protocatechuic	COOH	H	H	OH	OH	
Gallic	COOH	H	OH	OH	OH	
Gentisic	COOH	OH	H	H	OH	
Homogentisic	CH ₂ COOH	OH	H	H	OH	
Vanillic	COOH	H	OCH ₃	OH	H	
5-Sulphosalicylic	COOH	OH	H	H	HSO ₃	
Syringic	COOH	H	OCH ₃	OH	OCH ₃	
Veratric	COOH	H	OCH ₃	OCH ₃	H	
Vanillin	CHO*	H	OCH ₃	OH	H	

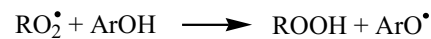
Fig. (3). Chemical structure of the benzoic acid derivatives found in mushrooms.

* Aldehydes are groups under the corresponding phenolic acid class.

Hydroxybenzoic acid derivatives commonly occur in the bound form and are typically a component of a complex structure like lignins and hydrolyzable tannins. They can also be found linked to sugars or organic acids in plant foods. Hydroxycinnamic acid derivatives are mainly present in the bound form, linked to cell-wall structural components, such as cellulose, lignin, and proteins, as well as associated to organic acids, such as tartaric or quinic acids (i.e., chlorogenic acids), through ester bonds [38].

The overall effectiveness of a natural phenolic antioxidant depends on the involvement of the phenolic hydrogen in radical reactions, the stability of the natural antioxidant radical formed during radical reactions, and the chemical substitutions present on the structure [99]. The substitutions on the structure are probably the most significant with respect to the ability of a natural antioxidant to participate in the control of

radical reactions and to form resonance-stabilized natural antioxidant radicals. The role of phenolic antioxidants (ArOH) is to interrupt the chain reaction according to:



To be effective ArO[•] must be a relatively stable free radical, so that it reacts slowly with substrate RH but rapidly with RO₂[•], hence the term “chain-breaking antioxidant” [100].

Like other phenolic compounds, the antioxidant activity of phenolic acids is due to the phenolic hydrogens. Hydroxyl substitutions at *ortho* and *para* positions also will enhance antioxidant activity. Intramolecular hydrogen bonds are formed by *ortho* substituted phenols (e.g., 1, 2-dihydroxybenzene) during radical reactions, which increase the stabil-

Cinnamic acid derivatives	Substitution				
	X	R ¹	R ²	R ³	R ⁴
<i>p</i> -Coumaric	H	H	H	OH	H
<i>o</i> -Coumaric	H	OH	H	H	H
Caffeic	H	H	OH	OH	H
Ferulic	H	H	CH ₃ O	OH	H
Sinapic	CH ₃ O	H	CH ₃ O	OH	CH ₃ O
3- <i>O</i> -caffeoylquinic	*	H	OH	OH	H
4- <i>O</i> -caffeoylquinic	*	H	OH	OH	H
5- <i>O</i> -caffeoylquinic	*	H	OH	OH	H

Fig. (4). Chemical structure of the cinnamic acid derivatives found in mushrooms.

* The carboxylic group is esterified with quinic acid.

ity of the phenoxy radical. It has been reported that caffeic acid is a better antioxidant than ferulic acid or *p*-coumaric acid. The second hydroxy group at *ortho* position allows the formation of intramolecular hydrogen bonds, which results in stronger antioxidant activity than those of compounds containing a methoxy (OCH₃) substitution *ortho* to the hydroxy group. Ferulic acid contains *ortho* methoxy substitution in its structure that may provide a stabilizing effect on the phenoxy radical, which enhances its antioxidant activity over *p*-coumaric acid. The higher antioxidant activity of trihydroxybenzoic acid (*i.e.*, gallic acid) over 3,4- dihydroxybenzoic acid (*i.e.*, protocatechuic acid) is due to the presence of three hydroxyl groups in trihydroxybenzoic acid [99, 101].

The acid proton appears to have little impact on antioxidant activity. Both caffeic acid and chlorogenic acid, the resultant compound after replacement of the acid proton of caffeic acid with quinic acid *via* an ester bond, were equally effective in controlling lipid oxidation [99]. The allylic group, as found in cinnamic acid derivatives, provides enhanced antioxidant activity when compared to benzoic acid derivatives. Caffeic acid (3,4- dihydroxycinnamic acid) was reported to be a better antioxidant than protocatechuic acid (3, 4-dihydroxybenzoic acid) in a lard system. The allylic group may improve the resonance stability of the phenoxyl radical [102].

In general, it is assumed that only plants possess the biosynthetic ability to produce flavonoids and not animals and fungi. Even though some flavonoids have exceptionally been reported from fungi *Aspergillus candidus* and *Phallus impudicus* [103] and more recently in mushrooms (Table 3). Recently, Barros *et al.* [88] reported that no flavonoids were detected in sixteen Portuguese wild mushrooms.

Flavonoids represent a large group of phenolic compounds with antioxidant activity, that occur naturally in plants and are found in fruits, vegetables, grains, barks, roots, stems, flowers, and derived products like tea and wine. These compounds have been linked to reduce the risk of major chronic diseases [38]. They are characterized for the carbon skeleton C6–C3–C6. The basic structure of these compounds consists of two aromatic rings (A and B rings) linked by a three carbon chain that is usually in an oxygenated heterocycle ring, or C ring (Fig. 5) [103].

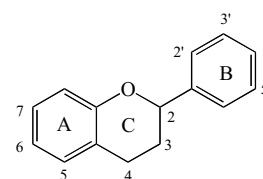


Fig. (5). The generic structure of flavonoids.

Several classes of flavonoids are delineated on the basis of differences in the generic structure of the heterocycle C ring and can be classified into flavonols, flavones, flavanols, flavanones, anthocyanins and isoflavonoids [104]. Fig. (6) presents the chemical structure of the flavonoids identified in wild mushrooms.

Flavonoids are most frequently found in nature as conjugates in glycosylated or esterified forms but can also occur in food as aglycones, especially as a result of the effects of food processing. Flavonols are the most abundant flavonoids in foods [105].

Multiple mechanisms have been identified as involved in the health-promoting effects of flavonoids, including antioxidant, anti-inflammatory and anti-proliferative activities, inhibition of bioactivating enzymes, or induction of detoxifying enzymes [106]. The antioxidant property of flavonoids was the first mechanism of action studied, in particular with regard to their protective effect against cardiovascular diseases. Flavonoids have been shown to be highly effective scavengers of most types of oxidizing molecules, including singlet oxygen and various free radicals, which are possibly involved in DNA damage and tumor promotion [100].

Phenolic compounds have specific health effects, even though they are non-nutritive compounds. In our diet they might provide health benefits associated with reduced risk of chronic diseases that may be due to their ability to reduce agents by donating hydrogen and quenching singlet oxygen. Antioxidant properties of phenolic compounds also play a vital role in the stability of food products, as well as in the antioxidative defence mechanisms of biological systems [107].

Tocopherols

Vitamin E is a term frequently used to designate a family of chemically related compounds, namely tocopherols and

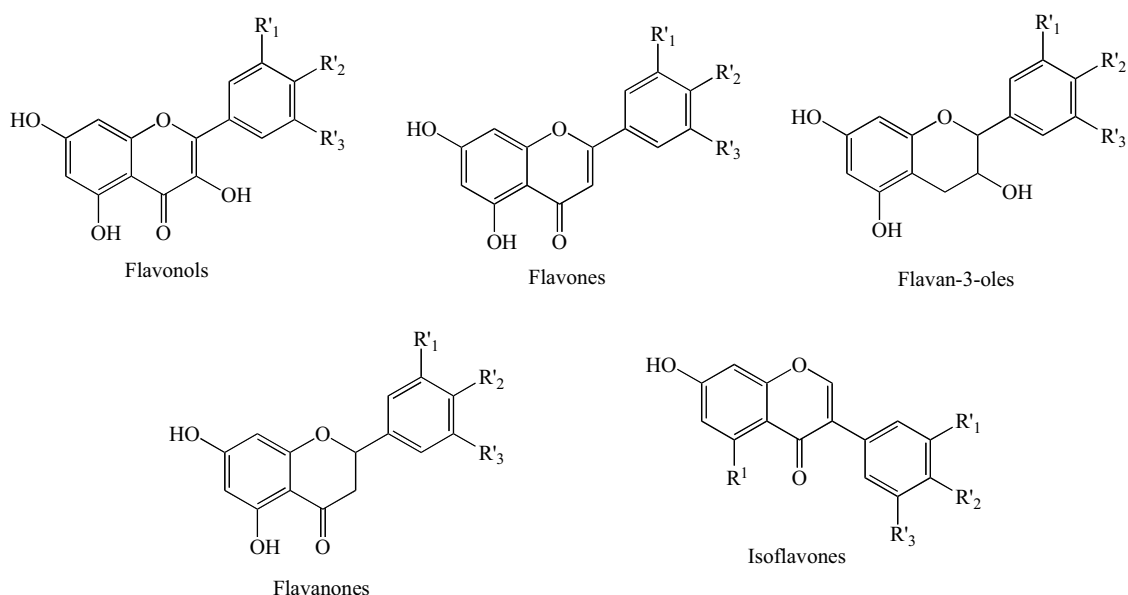


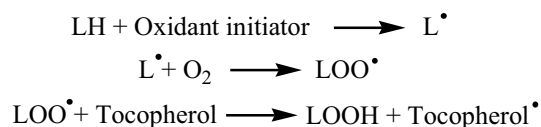
Fig. (6). Chemical structures of several flavonoids found in wild mushrooms. Flavonols: Quercetin ($R_1=R_2=OH$, $R_3=H$), Rutin ($R_1=R_2=OH$, $R_3=H$; OH in position-3 is substituted with the disaccharide rutinose), Kaempferol ($R_1=R_3=H$, $R_2=OH$), Myricetin ($R_1=R_2=R_3=OH$). Flavones: Chrysin ($R_1=R_2=R_3=H$). Flavan-3-oles: Catechin ($R_1=H$, $R_2=R_3=OH$). Flavanones: Hesperetin ($R_1=OH$, $R_2=R_3=H$), Naringenin ($R_1=R_3=H$, $R_2=OH$), Naringin ($R_1=R_3=H$, $R_2=OH$; OH in position-7 is substituted with the disaccharide rutinose). Isoflavones: Formononetin ($R^1=H$, $R_1=R_3=H$, $R_2=OCH_3$), Biochanin ($R^1=OH$, $R_1=R_3=H$, $R_2=OCH_3$).

tocotrienols, which share a common structure with a chromanol head and isoprenic side chain. Vitamin E is composed of eight chemical compounds: α -, β -, γ - and δ - tocopherols and four corresponding tocotrienols (Fig. 7) [108]. It is an important natural antioxidant in foods, especially those rich in polyunsaturated fatty acids. Due to its role as a scavenger of free radicals, vitamin E is also believed to protect our bodies against degenerative malfunctions, mainly cancer and cardiovascular diseases [109, 110].

Some reports have been published on the tocopherols content of mushrooms (Table 4). All reported the same methodology including saponification in the extraction process and analysis by HPLC coupled to UV detector. Only Barros *et al.* [87] described an extraction process without saponification, adding an antioxidant to avoid tocopherols oxidation, using special precautions to protect the samples from light and heat, and a fluorescence detector. α -, β -, δ -

and γ -Tocopherols were identified and quantified in wild mushrooms, while tocotrienols were not detected in any of the cited studies.

Vitamin E reacts with peroxy radicals produced from polyunsaturated fatty acids in membrane phospholipids or lipoproteins to yield a stable lipid hydroperoxide. They act as antioxidants by donating a hydrogen atom to peroxy radicals of unsaturated lipid molecules, forming a hydroperoxide and a tocopheroxyl radical, which reacts with other peroxy or tocopheroxyl radicals forming more stable adducts [111].



In the past α -tocopherol was considered the most active form of vitamin E in humans and it was reported to exhibit

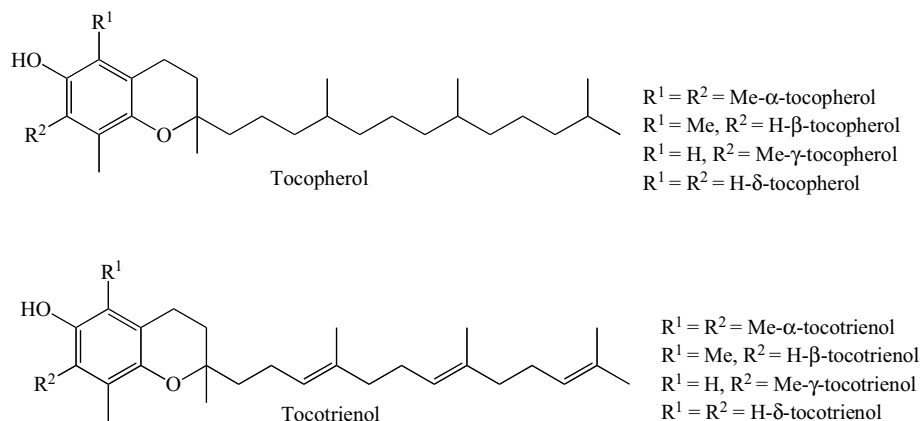


Fig. (7). Chemical structure of tocopherols and tocotrienols.

Table 4. Tocopherols Detected in Wild Mushrooms

Tocopherol	Mushroom species	Country	Ref.
α-tocopherol	<i>Auricularia mesenterica</i> , <i>Auricularia polytricha</i> , <i>Auricularia fuscusuccinea</i> (brown), <i>Auricularia fuscusuccinea</i> (white), <i>Tremella fuciformis</i>	Taiwan	[93]
	<i>Grifola frondosa</i> , <i>Morchella esculenta</i> , <i>Termitomyces albuminosus</i>	Taiwan	[57]
	<i>Agaricus bisporus</i> , <i>Polyporus squamosus</i> , <i>Pleurotus ostreatus</i> , <i>Lepista nuda</i> , <i>Russula delica</i> , <i>Boletus badius</i> , <i>Verpa conica</i>	Turkey	[70]
	<i>Hypsizigus marmoreus</i>	Taiwan	[73]
	<i>Agaricus blazei</i> , <i>Agrocybe cylindracea</i> , <i>Boletus edulis</i>	Taiwan	[76]
	<i>Agaricus arvensis</i> , <i>Agaricus bisporus</i> , <i>Agaricus silvaticus</i> , <i>Agaricus silvicola</i>	Portugal	[87]
	<i>Boletus edulis</i> , <i>Calocybe gambosa</i> , <i>Cantharellus cibarius</i> , <i>Craterellus cornucopioides</i> , <i>Marasmius oreades</i>	Portugal	[94]
	<i>Hypholoma fasciculare</i> , <i>Lepista nuda</i> , <i>Lycoperdon molle</i> , <i>Lycoperdon perlatum</i> , <i>Ramaria botrytis</i> , <i>Tricholoma acerbum</i>	Portugal	[79]
	<i>Pleurotus ostreatus</i>	India	[81]
	<i>Hypsizigus marmoreus</i>	Taiwan	[48]
β- tocopherol	<i>Agaricus arvensis</i> , <i>Agaricus bisporus</i> , <i>Agaricus romagnesii</i> , <i>Agaricus silvaticus</i> , <i>Agaricus silvicola</i>	Portugal	[87]
	<i>Boletus edulis</i> , <i>Calocybe gambosa</i> , <i>Cantharellus cibarius</i> , <i>Craterellus cornucopioides</i> , <i>Marasmius oreades</i>	Portugal	[94]
	<i>Hypholoma fasciculare</i> , <i>Lepista nuda</i> , <i>Ramaria botrytis</i> , <i>Tricholoma acerbum</i>	Portugal	[79]
δ- tocopherol	<i>Auricularia polytricha</i> , <i>Auricularia fuscusuccinea</i> (brown), <i>Tremella fuciformis</i>	Taiwan	[93]
	<i>Hericium erinaceus</i>	Taiwan	[50]
	<i>Grifola frondosa</i> , <i>Morchella esculenta</i> , <i>Termitomyces albuminosus</i>	Taiwan	[57]
	<i>Hypsizigus marmoreus</i>	Taiwan	[73]
	<i>Agaricus blazei</i> , <i>Agrocybe cylindracea</i> , <i>Boletus edulis</i>	Taiwan	[76]
	<i>Hypsizigus marmoreus</i>	Taiwan	[48]
γ- tocopherol	<i>Auricularia mesenterica</i> , <i>Auricularia polytricha</i> , <i>Auricularia fuscusuccinea</i> (brown), <i>Auricularia fuscusuccinea</i> (white), <i>Tremella fuciformis</i>	Taiwan	[93]
	<i>Ganoderma lucidum</i> , <i>Ganoderma tsugae</i>	Taiwan	[49]
	<i>Grifola frondosa</i> , <i>Tricholoma giganteum</i>	Taiwan	[50]
	<i>Lentinula edodes</i> , <i>Pleurotus cystidiosus</i> , <i>Pleurotus ostreatus</i>	Taiwan	[52]
	<i>Grifola frondosa</i> , <i>Morchella esculenta</i> , <i>Termitomyces albuminosus</i>	Taiwan	[57]
	<i>Agaricus blazei</i> , <i>Agrocybe cylindracea</i> , <i>Boletus edulis</i>	Taiwan	[76]
	<i>Boletus edulis</i> , <i>Calocybe gambosa</i> , <i>Cantharellus cibarius</i> , <i>Craterellus cornucopioides</i> , <i>Marasmius oreades</i>	Portugal	[94]
	<i>Hypholoma fasciculare</i> , <i>Lepista nuda</i> , <i>Tricholoma acerbum</i>	Portugal	[79]
	<i>Hypsizigus marmoreus</i>	Taiwan	[48]

the highest biological activity. However, many recent publications have been focused on the health effects of the other vitamin E isoforms [112, 113].

Ascorbic Acid

Ascorbic acid, also known as vitamin C, is related to the C6 sugars, being the aldono-1,4-lactone of a hexonic acid (L-galactonic or L-gulonic acid), and contains an enediol group on carbons 2 and 3 (Fig. 8); exists in two stereoisomeric forms: a L-isomer and a D-isomer [114]. It is a necessary nutrient for a limited number of animals, including humans, that are incapable of its synthesis and that must secure vitamin C by means of dietary uptake [115, 116]. In addition, ascorbic acid is thought to exert a protective role against various oxidative stress-related diseases such as heart dis-

ease, stroke, cancer, several neurodegenerative diseases and cataractogenesis [12].

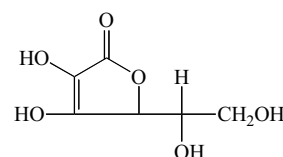


Fig. (8). Chemical structure of ascorbic acid.

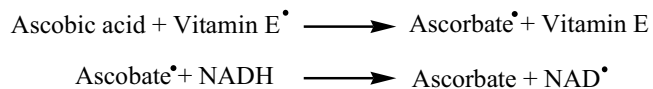
Vitamin C, one of the simplest vitamins, was found in several mushroom species (Table 5), either using HPLC coupled to UV or fluorescence detector, or following the spectrophotometer procedure based on the reaction with 2,6-dichlorophenolindophenol [117].

Table 5. Studies Reporting the Presence of Ascorbic Acid in Wild Mushrooms

Methodology	Mushroom species	Country	Ref.
HPLC-Fluorescence	<i>Agaricus bisporus</i> (white), <i>Agaricus bisporus</i> (brown), <i>Lentinus edodes</i> , <i>Pleurotus ostreatus</i>	Finland	[89]
Spectrophotometry	<i>Auricularia mesenterica</i> , <i>Auricularia polytricha</i> , <i>Auricularia fuscusuccinea</i> (brown), <i>Auricularia fuscusuccinea</i> (white), <i>Tremella fuciformis</i>	Taiwan	[93]
Spectrophotometry	<i>Grifola frondosa</i> , <i>Morchella esculenta</i> , <i>Termitomyces albuminosus</i>	Taiwan	[57]
Spectrophotometry	<i>Calvatia gigantea</i> , <i>Cantharellus cibarius</i> , <i>Russula integra</i> , <i>Gomphus floccosus</i> , <i>Lactarius quieticolor</i> , <i>Clavulina cinerea</i> , <i>Ramaria brevispora</i>	India	[95]
HPLC-UV	<i>Lactarius deliciosus</i> , <i>Suillus collinitus</i> , <i>Amanita caesarea</i>	Portugal	[91]
HPLC-UV	<i>Cantharellus cibarius</i>	Portugal	[85]
HPLC-UV	<i>Amanita rubescens</i> , <i>Suillus granulatus</i>	Portugal	[65]
Spectrophotometry	<i>Leucopaxillus giganteus</i> , <i>Sarcodon imbricatus</i>	Portugal	[67]
Spectrophotometry	<i>Lactarius piperatus</i>	Portugal	[68]
Spectrophotometry	<i>Lactarius deliciosus</i>	Portugal	[86]
Spectrophotometry	<i>Hypsizigus marmoreus</i>	Taiwan	[73]
HPLC-UV	<i>Fistulina hepatica</i>	Portugal	[92]
Spectrophotometry	<i>Agaricus arvensis</i> , <i>Agaricus bisporus</i> , <i>Agaricus romagnesii</i> , <i>Agaricus silvaticus</i> , <i>Agaricus silvicola</i>	Portugal	[78]
Spectrophotometry	<i>Hypoholoma fasciculare</i> , <i>Lepista nuda</i> , <i>Lycoperdon molle</i> , <i>Lycoperdon perlatum</i> , <i>Ramaria botrytis</i> and <i>Tricholoma acerbum</i>	Portugal	[79]
Spectrophotometry	<i>Boletus edulis</i> , <i>Calocybe gambosa</i> , <i>Cantharellus cibarius</i> , <i>Craterellus cornucopioides</i> , <i>Marasmius oreades</i>	Portugal	[94]
Spectrophotometry	<i>Agaricus bisporus</i>	Poland	[90]
HPLC-UV	<i>Pleurotus ostreatus</i>	India	[81]

Vitamin C can protect biomembranes against lipid peroxidation damage by eliminating peroxy radicals in the aqueous phase before the latter can initiate lipid peroxidation [114]. Vitamin C is effective against superoxide, hydroxyl radical, hydrogen peroxide, peroxy radical and singlet oxygen [118]. Cooperative interactions exist among vitamin C and vitamin E. They interact synergistically at the membrane-cytosol interface to regenerate membrane-bound oxidized vitamin E [119] (Fig. 2). As a lipophilic antioxidant, vitamin E can interact with the lipid components in the vascular systems, notably LDL, and protects them from atherogenic oxidative modification [120]. Conversely, the lipid-bound α -tocopherols can be oxidized by aqueous-phase radicals and transformed into reactive tocopherol radicals, which, in turn, react with the unsaturated lipids of the lipoprotein, initiating lipid oxidation by a tocopherol-mediated peroxidation reaction. Oxidized vitamin E can be reduced back to its antioxidant form by other aqueous-phase reductants, like ascorbic acid [121].

Ascorbic acid reacts rapidly with the tocopherol radical by reducing the ascorbate radical (semidehydroascorbate) to ascorbate by NADH-dependent semidehydroascorbate reductase.



In addition, ascorbate may sequester aqueous radicals in the plasma before they can oxidize vitamin E in the lipid

phase and affords protection for lipid-bound tocopherols. The interactions among these antioxidant nutrients are likely very important in protecting cells because the concentration of each antioxidant alone may not be adequate to effectively protect these cells against lipid peroxidation [122, 123].

Carotenoids

Carotenoids are nature's most widespread pigments and have also received substantial attention because of both their provitamin and antioxidant roles [38]. Particularly, β -carotene was found in several mushroom species (Table 6) either using HPLC coupled to UV or fluorescence detector, or following spectrophotometer procedures.

Carotenoids are synthesized by plants and microorganisms but not animals. Fruits and vegetables constitute the major sources of carotenoids in human diet [124, 125]. Close to 90% of the carotenoids in the diet and human body are represented by β -carotene, α -carotene, lycopene, lutein and β -cryptoxanthin [126] (Fig. 9).

All the carotenoids possess a 40-carbon skeleton of polyisoprenoid structure, a long conjugated double bonds chain forming the central part of the molecule and a near bilateral symmetry around the central double bond, as common chemical features [127]. This gives them their shape, chemical reactivity, and light-absorbing properties. Different carotenoids are derived essentially by modifications in the base structure by cyclization of the end groups and by introduction of oxygen-containing functional groups, giving them

Table 6. Studies Reporting the Presence of β -Carotene in Wild Mushrooms

Methodology	Mushroom species	Country	Ref.
Spectrophotometry	<i>Leucopaxillus giganteus, Sarcodon imbricatus</i>	Portugal	[67]
Spectrophotometry	<i>Lactarius piperatus</i>	Portugal	[68]
Spectrophotometry	<i>Lactarius deliciosus</i>	Portugal	[86]
HPLC-UV	<i>Agaricus bisporus, Polyporus squamosus, Lepista nuda, Russula delica, Verpa conica</i>	Turkey	[70]
Spectrophotometry	<i>Agaricus arvensis, Agaricus bisporus, Agaricus romagnesii, Agaricus silvaticus, Agaricus silvicola</i>	Portugal	[78]
Spectrophotometry	<i>Hypoholoma fasciculare, Lepista nuda, Lycoperdon molle, Lycoperdon perlatum, Ramaria botrytis and Tricholoma acerbum</i>	Portugal	[79]
Spectrophotometry	<i>Boletus edulis, Calocybe gambosa, Cantharellus cibarius, Craterellus cornucopioides, Marasmius oreades</i>	Portugal	[94]
HPLC-UV	<i>Pleurotus ostreatus</i>	India	[81]
HPLC-UV	<i>Hypsizigus marmoreus</i>	Taiwan	[73]
HPLC-UV	<i>Hypsizigus marmoreus</i>	Taiwan	[48]

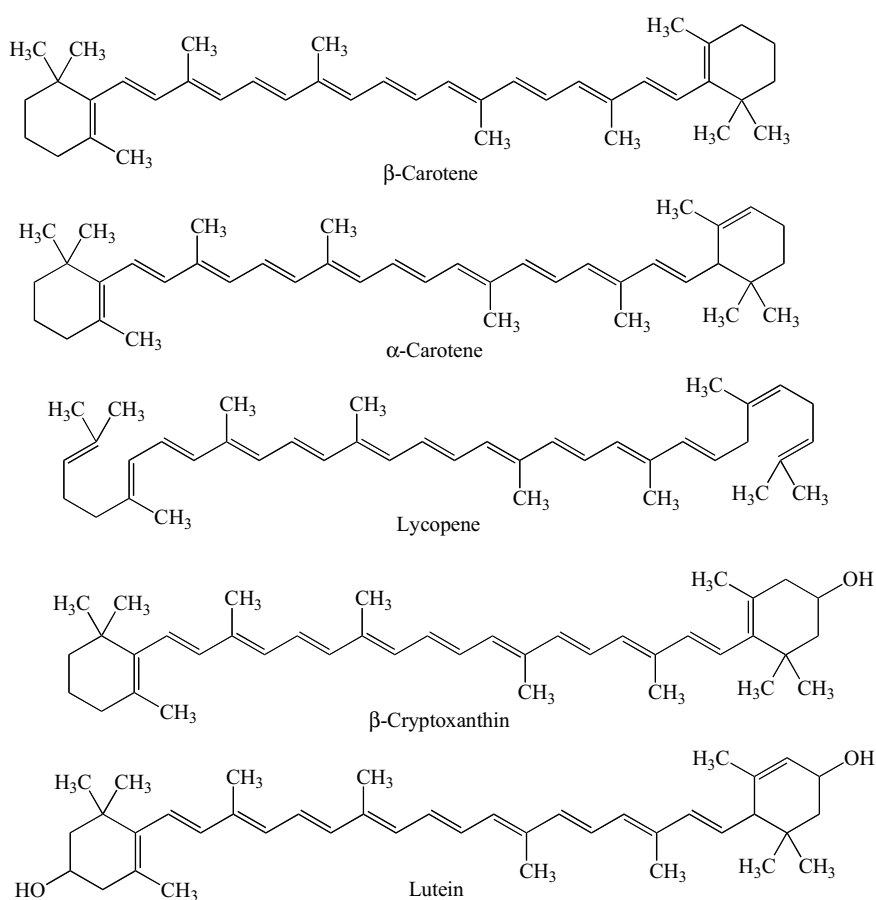


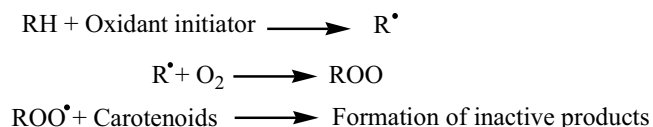
Fig. (9). Structures of some major dietary carotenoids.

their characteristic colors and antioxidant properties. Lycopene and β -carotene are examples of acyclized and cyclized carotenoids, respectively. Due to the presence of the conjugated double bonds, carotenoids can undergo isomerization to *cis-trans* isomers. Although the *trans* isomers are more common in foods and are more stable, very little is known about the biological significance of carotenoid isomerization in human health [128]. Carotenoids are thought to be respon-

sible for the beneficial properties in preventing human diseases including cardiovascular diseases, cancer and other chronic diseases [129]. They are important dietary sources of vitamin A, being β -Carotene, α -carotene, and β -cryptoxanthin able to function as provitamin A [38].

In recent years their antioxidant properties have been the major focus of research. Carotenoids can react with free radicals and become radicals themselves. They function as a

chain-breaking antioxidant in a lipid environment, especially under low oxygen partial pressures [128, 130]. The peroxy radicals (ROO[•]) formed from lipids (especially polyunsaturated phospholipids) are very damaging to cells. The extensive systems of double bonds make carotenoids susceptible to attack by peroxy radicals, resulting in the formation of inactive products [122].



Carotenoids reactivity depends on the length of the chain of conjugated double bonds and the characteristics of the end groups. Carotenoid radicals are stable by virtue of the delocalization of the unpaired electron over the conjugated polyene chain of the molecules. This delocalization also allows addition reactions to occur at many sites on the radical [127].

Overall, wild mushrooms contain different antioxidants such as phenolic compounds, tocopherols, ascorbic acid, and carotenoids which could be extracted for the purpose of being used as functional ingredients namely against chronic diseases related to oxidative stress. Also, mushrooms might be used directly in diet and promote health, taking advantage of the additive and synergistic effects of all the bioactive compounds present.

Public health authorities consider prevention with nutraceuticals as a powerful instrument in maintaining and promoting health, longevity and life quality.

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