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Roots and rhizomes of wild *Asparagus*: Nutritional composition, bioactivity and nanoencapsulation of the most potent extract

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ABSTRACT

The nutritional composition and bioactive properties of roots and rhizomes of *Asparagus stipularis* were evaluated. Antioxidant activity of extracts obtained by infusion was evaluated using free radicals scavenging and reducing power methods. Porcine liver primary cell was used to check the hepatotoxicity of infusions. Results revealed that *Asparagus* samples are likely a source of nutrients, such as dietary fibre and essential fatty acids. HPLC-DAD-ESI/MS characterization of infusions allowed the identification and quantitation of 7 phenolic compounds, all hydroxycinnamoyl derivatives, with caffeic acid as the most abundant. Roots infusion contained the highest amounts of these compounds. It also exhibited the highest antioxidant activity in all assays, with EC₅₀ values of 0.44 ± 0.01, 0.98 ± 0.03 and 0.64 ± 0.01 mg/mL for DPPH, ABTS and FRAP assays, respectively, with no toxicity towards PLP2 primary cell cultures (GI₅₀ > 400 µg/mL). PLGA nanoparticles loaded with root extract were prepared using solvent-evaporation double emulsion method. Nanoparticles size was about 260 nm and a polydispersity index around 0.1, with a zeta potential of about -36 mV, as well as a good encapsulation efficiency of approximately 83%. Their morphology was analysed by SEM and spherical polymeric nanoparticles with a smooth surface were observed. FTIR and DSC were also performed, which allowed corroborating the efficacy of the encapsulation and to confirm the production of a stable and robust system to load *Asparagus* extracts. The developed nanoparticles are expected to be used as delivery systems for bioactive compounds of *A. stipularis* and they could be used as an innovative dietary supplement.

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1. Introduction

Asparagus is a genus that belongs to the Liliaceae family and contains over 300 species widely diffused in the world continents (Kanno & Yokoyama, 2011). Eastern civilizations have used *Asparagus* extracts as laxatives, stimulants, and diuretics for centuries (Fuentes-Alventosa et al., 2009). Furthermore, Greek medicine used *Asparagus* extracts as a tonic to cure many ailments, such as the rheumatic, cancer, rheumatic, liver, and kidney diseases (Pegiou et al., 2019). The genus *Asparagus* includes commercially important species, essentially *A. officinalis* L. and *A. albus* L., whereas the remaining species are poorly known, such as *A. stipularis* Forssk (Serairi-Beji et al., 2017). In Spain, this latter plant has been traditionally used to deal with inflammatory diseases (Bremner et al., 2009). Because of its high iron, nitrogen, and ascorbic acid contents (Ulukapi et al., 2014), *A. stipularis* can substitute other *Asparagus* species for human nutrition. A triterpene saponin (asparagalin A) and two alkaloids (5-hydroxyaspartipuline and aspartipuline) have been purified from the roots of *A. stipularis* (El-Seedi et al., 2012; Galala et al., 2015). Unfortunately, plant extract often present some problems, including long-term instability during storage, as influenced by temperature, light, pH and oxygen conditions, and low nutrient and phytochemical bioavailability (Coimbra et al., 2011; Fang & Bhandari, 2010; Woranuch & Yoksana, 2013). The encapsulation of plant extracts into nanoparticles may reduce the outlined limitations and increase their functional properties (Lu et al., 2016). The polymer PLGA (poly-D, L-lactide-co-glycolide) was widely employed by the food and pharmaceutical industries to protect active ingredients and their controlled release (Stevanovic & Uskokovic, 2009). PLGA is recognized by the European Medicines Agency (EMA) and the US Food and Drug Administration (FDA), because of its biodegradability and low toxicity (Danhier et al., 2012), and has been previously used to encapsulate plant extracts. Suman and Gupta (2013) encapsulated roots extract of *Clerodendrum infortunatum* using PLGA to treat hypercholesterolemia, with an encapsulation efficiency of 98.40%. In another study, Ribeiro et al. (2015) developed PLGA nanoparticles loaded with an extract from *Uncaria tomentosa*, accomplishing significant antiproliferative properties against different cancer cell lines, including leukemia, breast cancer, melanoma, lung and colon carcinoma cells.

In the current study, we intended to explore the nutritional composition and bioactive compounds of *A. stipularis* roots and rhizomes and to assess its antioxidant potential and its cytotoxicity in non-tumor cells. In addition, since the extracts derived from *A. stipularis* infusion has not been encapsulated into nanoparticles so far, the encapsulation of the most active extract into PLGA nanoparticles was performed for future food, pharmaceutical, and cosmetic applications.

2. Materials and methods

2.1. Reagents and standards

Folin-Ciocalteu reagent (2 M), 2,2-Diphenyl-1-picrylhydrazyl (DPPH, free radical, 95%), phenol, gallic acid (98%), standards of tocopherols (α -, β -, γ - and δ -isoforms), organic acids (malic, oxalic and citric acids), fetal bovine serum (FBS), Dulbecco's modified Eagle's medium (DMEM), penicillin, streptomycin, non-essential amino acids, dichloromethane and polyvinyl alcohol (PVA) were bought from Sigma-Aldrich (St. Louis, MO, USA). Chlorotrimethylsilane were obtained from Acros Organics (New Jersey, USA) and hexamethylDisilazane is from MP Biomedicals, Germany. Carbohydrate standards used for identification were purchased from Sigma Chemical Co. (St. Louis, USA). Iron (III) chloride hexahydrate ($\text{FeCl}_3 \cdot 6\text{H}_2\text{O}$) and sodium acetate were obtained from Acros Organics (Geel, Belgium). Metals (Fe, Mn, Zn, and Cu) and 2, 2'-Azino-bis-(3-ethylbenzothiazoline-6-sulfonic acid potassium persulfate were purchased from Merck (Darmstadt, Germany). Phenolic compounds standards were purchased from Extrasynthese (Genay, France). PLGA (at ratio lactide to glycolide 50:50) was provided by

Corbion Purac (Amsterdam, The Netherlands). The distilled and bi-distilled water was prepared in-house (CBIOS, Lisbon, Portugal).

2.2. Plant material

Roots and rhizomes of wild *Asparagus stipularis* Forssk were harvested in Monastir (Tunisia). Samples were cut into small pieces and dried by active ventilation at 37 °C for 24–48 h using an air oven (BOV-T270C, Biobase, China). Dried samples were ground in an ultra-centrifugal mill ZM 200 (Retsch, Haan, Germany) and put into storage at -20 °C.

2.3. Nutritional value of *A. stipularis* roots and rhizomes

2.3.1. Proximate composition

The standard analytical methods described by the Association of Official Analytical Chemists (AOAC, 2005) were used to determine the moisture, fat, protein, and ash contents. The moisture content was obtained after drying 1 g of the samples in an air oven (105 °C) (BOV-T270C, Biobase, China). Soxhlet (SER 148 Solvent Extractor, Velp Scientifica, Milan, Italy) and Kjeldahl apparatus (model Pro-Nitro M Kjeldahl Steam Distillation System, Barcelona, Spain) were used to determine the fat and protein content, respectively. The content of ash was analysed by incineration at 550 ± 10 °C (Muffle Furnace mLs1200, Thermo Scientific, Monroe, LA, USA). The results were expressed as g/100 g of fresh weight (fw).

2.3.2. Available carbohydrates: soluble sugars, oligosaccharides and polyols

Soluble sugars, oligosaccharides and polyols were extracted following the method previously reported by Mechri et al. (2015). Analysis was performed using a Hewlett-Packard 5890 series II gas chromatograph equipped with an HP-5MS capillary column (30 m × 0.25 mm) and a flame ionization detection (FID) system. Individual compounds were identified using the relative retention times in comparison to that of the standards. The internal standard calculation method was adopted to quantify each compound. The analysis was conducted in triplicate, and results were expressed in mg/100 g fw.

2.3.3. Dietary fiber (soluble and insoluble)

The enzymatic-gravimetric methods (AOAC 993.19 and 991.42) were used to determine the fiber content (García-Herrera et al., 2014). Results of total dietary fiber, as well as soluble and insoluble dietary fiber were expressed as g/100 g fw.

2.3.4. Energy value

The energy value was determined using the following equation (European Parliament & Council of the European Union, 2011):

$$\text{Energy (Kcal / 100 g fresh weight)} = [4 \times (\text{g}_{\text{protein}} + \text{g}_{\text{soluble sugar}}) + 2 \times (\text{g}_{\text{fiber}}) + 9 \times (\text{g}_{\text{fat}})]$$

2.3.5. Mineral elements content

The mineral elements content was determined following the method 930.05 of AOAC (2005). Copper (Cu), Manganese (Mn), Iron (Fe) and Zinc (Zn), as microelements, were directly quantified by atomic absorption spectroscopy (AAS) (Analyst 200 PerkinElmer equipment (PerkinElmer, Waltham, MA, USA)). For macroelements, Magnesium (Mg), Calcium (Ca), Potassium (K), and Sodium (Na), an additional 1/10 (mL/mL) dilution of the sample and standards was added to prevent interferences between different elements. The results were expressed in mg/100 g fw.

2.3.6. Fatty acids profile and tocopherols contents

Fatty acids (FA) were evaluated using the GC method described by [Dhibi et al. \(2010\)](#), with some modifications. FA were analysed using a Hewlett-Packard gas chromatograph (Hewlett-Packard, Palo Alto, CA, USA) provided with a flame ionization detector and a *split-splitless* injector, set at 270 °C. The identification of FA was performed by comparison of their retention time to pure standards analysed under the same conditions. HP Chemstation integrator was used to calculate the FA peak areas and they were recorded as peak area percentages. Analysis were conducted in triplicate and the results were expressed as relative percentages.

Tocopherols profile was estimated following the procedure adopted by [Dias et al. \(2013\)](#). An HPLC system (Knauer, Smartline system 1000, Berlin, Germany) coupled to a fluorescence detector (FP-2020; Jasco, Easton, MD, USA) programmed for excitation at 290 nm and emission at 330 nm was used. Identification of individual tocopherols was performed using the method of chromatographic comparison with authentic standards. Quantification of tocopherols was performed using a calibration curve obtained from the commercial standards of each compound. Analysis was conducted in triplicate and the results were expressed in mg/100 g fw.

2.4. Bioactive compounds

2.4.1. Organic acids

A Shimadzu 20 A series UFLC (Shimadzu Corporation, Kyoto, Japan) was used to analyse the organic acids following the method previously described by [Barros, Duenas, Carvalho, Ferreira, and Santos-Buelga \(2012\)](#). The quantification of organic acids was performed by comparing the area of their peaks to the calibration curves established from the standards of individual organic acid. The experiment was conducted in triplicate and the results were expressed in g/100 g fw.

2.4.2. Volatile compounds

The volatile composition of *A. stipularis* samples was performed following the method of [Jelled et al. \(2016\)](#). Solid Phase Micro-extraction (SPME) was performed followed by Gas-Chromatography coupled to Mass Spectrometry (GC-MS) analysis. The identification of volatile compounds was performed by comparison between their linear retention indices (LRI), library mass spectra as well as MS literature data.

2.5. Roots and rhizomes infusion extracts preparation and analysis

The roots and rhizomes infusions were prepared following the procedure previously reported by [Dias et al. \(2013\)](#). Powdered roots or rhizomes (1 g) were added to 200 mL of water (100 °C) and left for 5 min at room temperature. Then, the solutions were filtered, frozen and lyophilized.

2.5.1. Evaluation of bioactive compounds in the infusions

2.5.1.1. Organic acids. Dried extracts (10 mg) of roots and rhizomes infusions were re-dissolved in 1 mL of meta-phosphoric acid (4.5%), filtered (0.22 µm) and analysed as above described. The results were expressed in mg/g extract.

2.5.1.2. Total polyphenol (TPC) and total flavonoids (TFC) contents. The *Folin-Ciocalteu* reagent was used to evaluate the TPC of *A. stipularis* infusions ([Wolfe et al., 2003](#)). The absorbance was measured at 765 nm against a blank without extract using a UV-vis spectrophotometer (PerkinElmer Lambda 40 UV/VIS Spectrophotometer). The calculation was conducted using a gallic acid calibration curve and results were expressed as mg of gallic acid equivalents per gram of extract (mg GAE/g).

The colorimetric method adapted by [Jia et al. \(1999\)](#) was adopted to quantify the flavonoids. The absorbance was measured at 510 nm against the blank without extract. Results were expressed in mg of catechin equivalents (EC) per gram of extract.

2.5.1.3. HPLC-DAD-ESI/MS analysis for individual phenolic compounds.

The analysis of phenolic compounds contained in the extracts was performed by HPLC using double online detection by diode array spectrophotometry and mass spectrometry (MS) ([Adouni et al., 2018](#)). A Hewlett-Packard 1100 chromatography (Hewlett-Packard 1100, Agilent Technologies, Santa Clara, CA, USA) equipped with a quaternary pump and a diode array detector (DAD) controlled by a HP ChemStation data-processing station (rev. A.05.04) was used. The identification of individual phenols was performed by comparing their retention characteristics, UV and mass spectra to the available standards, literature data and our database library. The quantitative analysis was conducted using a calibration curve from caffeic acid ($y = 529.2x - 7.6$). The results were expressed in mg per 100 g of extract dry weight (dw).

2.6. *In vitro* biological activities of *A. stipularis* infusions extracts

2.6.1. Antioxidant activity assays

2.6.1.1. ABTS radical scavenging activity. An improved ABTS^{•+} method previously described by [Re et al. \(1999\)](#) was adopted. A mixture of 39.2 mg ABTS and 6.7 mg potassium persulfate was prepared and left in dark at room temperature for 15–16 h before use. The obtained ABTS^{•+} solution was diluted with ethanol to obtain the absorbance of 0.700 ± 0.02 at 734 nm. Later, 3.9 mL of ABTS^{•+} solution were mixed with each infusion and left to stand for 6 min. The absorbance was measured at 734 nm in a UV-visible spectrophotometer. Blank was absolute ethanol. The antioxidant activity was determined by calculating the percentage of inhibition of the ABTS^{•+} radical according to formula (1):

$$\% \text{ Inhibition} = (\text{Abs}_{\text{control}} - \text{Abs}_{\text{test}}) / \text{Abs}_{\text{control}} \times 100$$

$\text{Abs}_{\text{control}}$ is the absorption of the control containing ABTS^{•+} ($t = 0$ min) and Abs_{test} is the absorption of the tested infusion ($t = 6$ min). The concentration of the extract that provides 50% of antioxidant activity (EC₅₀) was calculated. Trolox was used as standard and analyses were carried out in triplicate.

2.6.1.2. DPPH radical scavenging activity. The [Brand-Williams et al. \(1995\)](#) method was used with some modifications. A 50 µL aliquot of sample was mixed with 2 mL of DPPH· methanolic solution (89.7 µM). The mixture was left in the dark at room temperature for 45 min and the absorption measured at 517 nm. The experiment was carried out in triplicate. The antioxidant activity was determined by formula (2):

$$\% \text{ Inhibition} = (\text{Abs}_{\text{control}} - \text{Abs}_{\text{test}}) / \text{Abs}_{\text{control}} \times 100$$

$\text{Abs}_{\text{control}}$ is the absorption of the control containing DPPH[•] and Abs_{test} is the absorption of the tested infusion. The EC₅₀ was calculated. Trolox was used as a standard.

2.6.1.3. Ferric reducing antioxidant power (FRAP). The [Benzie and Strain \(1996\)](#) method was used with some modifications. Each infusion (50 µL) was mixed with distilled water (150 µL) and FRAP reagent (1500 µL). The mixture was vortexed for 15 s and left in the dark (2 h, 37 °C). The absorbance was measured at 690 nm with a UV-vis spectrophotometer (PerkinElmer Lambda 40 UV/VIS Spectrophotometer). The concentration of the extract that gives 0.5 absorbance (EC₅₀) was calculated from the graph against the concentration of infusions. The analysis was carried out in triplicate. Trolox was used as a standard.

2.6.2. Cytotoxic activity

A porcine liver primary cell culture (PLP2) was used to perform a

preliminary study of *in vitro* toxicity of *A. stipularis* infusion following the methodology previously described (Guimarães et al., 2013). The results were expressed in GI₅₀ values, the concentration that inhibited 50% of the net cell growth. As positive control, ellipticine was used.

2.7. Extract encapsulation into nanoparticles

2.7.1. PLGA nanoparticles production

A modified solvent emulsification-evaporation method previously described by Fonte et al. (2014) was adopted to produce PLGA nanoparticles (NPs). Briefly, an organic phase was prepared by dissolving PLGA 50:50 (100 mg) in dichloromethane (2 mL). After that, 0.2 mL of roots infusion (IR) were added to the previous solution and sonicated for 30 s at 70% of amplitude using a Q125 Sonicator (QSonica Sonicators, Newtown, CT, USA). Later, the emulsion was rapidly poured into PVA (10 mL, 2%) and sonicated using the same conditions. Next, the emulsion was magnetically stirred for 3 h at room temperature to evaporate the organic solvent. Control unloaded PLGA NPs were produced following the same procedure.

2.7.2. Physicochemical properties of nanoparticles

A Coulter Nano-sizer Delsa™ Nano C from Beckman Coulter, Inc. (Brea, CA, USA) was used to determine the particle size, polydispersity index (PDI) and zeta potential (ZP) of the *Asparagus* loaded NPs. The samples were diluted with distilled water. The mean particle size and PDI of NPs were analysed by dynamic light scattering (DLS). The zeta potential was determined by electrophoretic mobility analysis. Experiments were performed in triplicate.

2.7.3. Determination of encapsulation efficiency (EE)

Total phenolic content analysis (TPC) using the Folin-Ciocalteu assay, as above described (section 2.5.1.2) permitted to determine the encapsulation efficiency (EE) of roots infusion in the PLGA NPs. The amount of TPC entrapped within the nanoparticles was evaluated by measuring the amount of unloaded free total phenolic content in the supernatant recovered after ultracentrifugation of the nanoparticle suspension at 15,000 rpm for 20 min. After centrifugation, the nanoparticles were re-suspended in bidistilled water, followed by freeze-drying (48 h, -50 °C, 400 mTorr) using a LABCONCO FreeZone 25® freeze dryer (Kansas City, MO, USA). The TPC was determined by ultraviolet-visible (UV-VIS) spectrophotometry at 765 nm.

EE was calculated according to formula (3):

$$EE\% = \frac{\text{Amount of TPC in the encapsulated extract} - \text{Amount of TPC in the supernatant}}{\text{Amount of TPC in the encapsulated extract}} \times 100$$

2.7.4. Scanning electron microscopy (SEM)

The morphology of NPs was analysed by SEM using a FEI Quanta 400 FEG SEM (FEI, Hillsboro, OR, USA). Briefly, the freeze-dried NPs were placed on metal stubs. Then, the NPs were vacuum-coated with a layer of Gold/Palladium for 60 s with a current of 15 mA.

2.7.5. Fourier transform infrared spectroscopy (FTIR)

The molecular characterization of roots infusion, PLGA, physical mixture (roots infusion and PLGA) and unloaded and loaded NPs was performed by FTIR. The analysis was conducted with a MB3000 FTIR spectrometer (ABB, Zurich, Switzerland) linked to a MIRacle single reflection attenuated total reflectance (ATR) accessory from PIKE Technologies (Madison, WI, USA). All spectra were registered in the

range of 600–4000 cm⁻¹ with 50 scans at a resolution of 4 cm⁻¹.

2.7.6. Differential scanning calorimetry (DSC)

The roots infusion, PLGA, physical mixture (roots infusion and PLGA) and unloaded and loaded NPs were analysed using a differential Scanning Calorimeter DSC 200 F3 Maia (Netzsch, Selb, Germany). Each sample (2 mg) was sealed in an aluminium pan. The heating curves were recorded using a heating rate of 10 °C/min from 20 °C to 260 °C. An empty pan was used as reference.

2.8. Statistical analysis

All the extractions and analyses were carried out in triplicate. Results are expressed as mean values ± standard deviation (SD), being analysed using a Student's t-test, with *p* = 0.05. The analysis was performed using Origin Pro 8 (Origin Lab).

3. Results and discussion

3.1. Nutritional value and bioactive compounds of *A. stipularis* roots and rhizomes

The results of the nutritional characterization of *A. stipularis* roots and rhizomes are presented in Table 1. They showed a similar proximate composition: moisture was the major component, followed by carbohydrates, protein, fat and ash. In our previous study on shoots of *A. stipularis* (Adouni et al., 2018), a higher moisture content was found (82.38%), with consequent lower percentages of the other fractions (0.82, 1.19 and 3.64 g per 100 g for ash, proteins and fat, respectively).

The major differences between the two samples were observed for dietary fiber and mineral contents. Indeed, the contents of total soluble (2.22 ± 0.14 g/100 g fw) and insoluble dietary fiber (30.88 ± 3.21 g/100 g fw) in the roots were two-fold higher than those in the rhizomes (1.40 ± 0.11 and 14.06 ± 1.71 g/100 g fw, respectively). In rhizomes, a difference around 19% existed between the amount of total carbohydrates and the sum of fiber and soluble sugars, suggesting that a part of carbohydrates was not included under those fractions, which should be explained by the presence of non-fiber oligo- or carbohydrates, such as starch. Dietary fiber has received much importance based on its numerous health benefits for instance in glycemic control in diabetes (Weickert et al., 2018), decreasing cholesterol levels and inflammation (Wannamethee et al., 2009; Muller et al., 2018). The current fiber intake prescribed by the Food and Nutrition Board ranges between 21 and 38

g/day for adults, depending on life stage groups (Trumbo et al., 2002). Thus, a portion of 100 g of rhizomes or roots of *A. stipularis* could cover at least 74 and 87% of this dietary recommendation, respectively. The recommended European consumption of fiber is estimated to be 20 g/person/day, so an increase in fiber consumption seems needed, a goal to which supplements of the studied samples could contribute.

The contents of essential minerals Na, Ca, K, Mg and trace minerals Zn, Fe, Mn and Cu of the *Asparagus* samples are also shown in Table 1. Among trace minerals, Fe was the main element in both samples. Iron is involved in many aspects of energy metabolism and is an important component in the structure and function of haemoglobin, myoglobin and cytochromes and a co-factor for many enzymes (Puntarulo, 2005). Recommended values for daily Fe intake are about 16–20 mg/day for women under 50–55 years old and 8–10 mg/day for men and elderly women (EFSA, 2006). Thus, a portion of 100 g of roots could cover at

Table 1Proximate composition, mineral elements and energy value of *Asparagus stipularis* roots and rhizomes.

| Proximate composition (g/100 g fw) | Roots | Rhizomes |
|--|---------------------------|---------------------------|
| Moisture | 57.4 ± 0.8 ^a | 59 ± 1 ^a |
| Crude proteins | 6.3 ± 0.2 ^b | 4.9 ± 0.2 ^a |
| Total Fat | 0.27 ± 0.04 ^b | 0.15 ± 0.03 ^a |
| Total ash | 1.99 ± 0.03 ^b | 1.44 ± 0.07 ^a |
| Total carbohydrates | 91.47 ± 0.15 ^b | 93.45 ± 0.27 ^a |
| Soluble sugars | 1.13 ± 0.01 ^b | 0.67 ± 0.01 ^a |
| Total dietary fiber | 33 ± 3 ^b | 15 ± 1 ^a |
| Soluble dietary fiber | 2.22 ± 0.14 ^b | 1.40 ± 0.11 ^a |
| Insoluble dietary fiber | 30.88 ± 3.21 ^b | 14.06 ± 1.71 ^a |
| Macro and microelements (mg/100 g fw) | | |
| Fe | 15.5 ± 0.2 ^b | 8.5 ± 0.3 ^a |
| Cu | 0.98 ± 0.01 ^b | 0.63 ± 0.03 ^a |
| Mn | 0.28 ± 0.01 ^b | 0.13 ± 0.05 ^a |
| Zn | 0.93 ± 0.03 ^a | 1.7 ± 0.02 ^b |
| Ca | 332 ± 6 ^b | 211 ± 2 ^a |
| Mg | 106 ± 4 ^a | 128.0 ± 0.3 ^b |
| Na | 12.0 ± 0.3 ^a | 14.3 ± 0.3 ^b |
| K | 161 ± 4 ^a | 278 ± 5 ^b |
| Energy (kcal/100 g fw) | 98 ± 6 ^b | 55 ± 4 ^a |

In each row different letters mean significant differences ($p < 0.05$).

least 77% of Recommended Dietary Allowance (RDA) for women and more than 100% of RDA for men. Considering the analysed essential minerals (Table 1), the rhizomes revealed higher contents of Na (14.3 ± 0.3 mg/100 g fw) and K (278 ± 5 mg/100 g fw) than roots (12 and 161 mg/100 g fw, respectively). K and Na control the ionic balance of the human body and sustain tissue excitability. Na/K ratios lower than 1, as existing in (most) vegetables contribute to ameliorate sodium-related health disorders and in particular hypertension. This ratio was maintained well below 1 in the studied samples as well as in the shoots of *A. stipularis* reported in our previous work (Adouni et al., 2018). Magnesium is a necessary mineral for the human body. It is required as a co-factor in many enzymes and it plays a role in nucleic acid synthesis. Intakes of 350 mg/day and 300 mg/day have been recommended by the EFSA for men and women, respectively (EFSA, 2015). A higher content of Mg was found for the rhizomes than for roots. Calcium was the essential mineral present in higher levels in *A. stipularis* roots, a notable concentration taking into account that the needs of this elements are situated around 1 g/day for adults and, but for dairy products, there is a scarcity of dietary sources. This element possesses great nutritional

Table 2Soluble sugars and organic acids composition in *Asparagus stipularis* roots and rhizomes.

| Soluble sugars (mg/100 g fw) | Roots | Rhizomes |
|-----------------------------------|--------------------------|--------------------------|
| Fructose | 26.5 ± 0.3 ^a | 81.4 ± 0.9 ^b |
| Glucose | 150 ± 2 ^b | 45.6 ± 0.3 ^a |
| Sucrose | 286 ± 4 ^a | 466 ± 5 ^b |
| Galactose | 56 ± 1 ^b | 13.4 ± 0.3 ^a |
| Rhamnose | 112 ± 3 ^b | 5.32 ± 0.03 ^a |
| Xylose | 39 ± 1 ^b | 2.59 ± 0.07 ^a |
| Arabinose | nd | 3.5 ± 0.1 |
| Raffinose | 124 ± 4 ^b | 19.5 ± 0.3 ^a |
| Trehalose | nd | 20.4 ± 0.3 |
| Sorbitol | 34.0 ± 0.6 ^b | 5.5 ± 0.3 ^a |
| Inositol | 304 ± 3 ^b | 9.40 ± 0.06 ^a |
| Organic acids (g/100 g fw) | | |
| Oxalic acid | 0.25 ± 0.03 ^b | 0.12 ± 0.03 ^a |
| Malic acid | tr | tr |
| Citric acid | 0.07 ± 0.01 ^a | 0.15 ± 0.04 ^b |
| Total organic acids | 0.33 ± 0.04 ^b | 0.27 ± 0.01 ^a |

In each row different letters mean significant differences ($p < 0.05$); nd: not detected; tr: traces.

interest because it is necessary for metabolic reactions, bone formation, teeth, muscle and heart function and also can act as activators for enzyme systems (FAO/WHO, 2001, pp. 223–224). Despite the differences found between roots and rhizomes in the concentrations of microelements and macroelements in the current study, in general, the *Asparagus* samples can be considered adequate sources of essential mineral elements that could be consumed within the current diet.

The content in total soluble sugars of *A. stipularis* roots (1.13 g/100 g fw) was higher than in rhizomes (0.67 g/100 g fw) (Table 2). Arabinose and trehalose were detected only in rhizomes, which also presented higher contents in fructose and especially sucrose, while the remaining sugars are found in greater concentrations in roots, namely glucose, galactose, rhamnose, xylose, raffinose, sorbitol and inositol.

The organic acid profiles are summarized in Table 2. Oxalic and citric acid were quantified, while malic acid was detected in trace amounts in both samples. Those organic acids were also detected in spears of *A. stipularis*, reported in our previous work (Adouni et al., 2018). The roots showed two-fold higher concentration of oxalic acid than the rhizomes, while the opposite happened with citric acid. Actually, rhizomes of *A. stipularis* revealed a citric acid content higher than other previously analysed wild edible plants, such as bulbs and pseudo stems of *Allium ampeloprasum* (Sánchez-Mata et al., 2012), fruits of *Arbutus unedo* (Pereira et al., 2013) or leaves of *Anchusa azurea* (Morales et al.,

Table 3Distribution of main fatty acid composition (relative percentage on total fatty acids) and tocopherols (mg/100 g fw) in *Asparagus stipularis* roots and rhizomes.

| Fatty acids | Relative percentage (%) | |
|----------------------------------|----------------------------|---------------------------|
| Saturated | Roots | Rhizomes |
| C6:0 | 0.37 ± 0.01 ^a | 0.36 ± 0.02 ^a |
| C12:0 | 2.14 ± 0.05 ^a | 2.27 ± 0.03 ^b |
| C14:0 | 0.14 ± 0.04 ^a | 0.13 ± 0.01 ^a |
| C15:0 | 0.81 ± 0.04 ^b | 0.51 ± 0.03 ^a |
| C16:0 | 22.82 ± 0.1 ^a | 24.10 ± 0.07 ^b |
| C17:0 | 1.4 ± 0.2 ^b | 1.03 ± 0.04 ^a |
| C18:0 | 8.21 ± 0.05 ^a | 8.57 ± 0.06 ^b |
| C20:0 | 3.7 ± 0.2 ^a | 3.99 ± 0.05 ^b |
| C22:0 | 0.57 ± 0.1 ^a | 0.70 ± 0.01 ^b |
| C23:0 | 0.79 ± 0.02 ^a | 0.80 ± 0.05 ^a |
| C24:0 | 0.16 ± 0.01 ^a | 0.15 ± 0.02 ^a |
| Monounsaturated | | |
| C16:1 (n-9) | 6.4 ± 0.1 ^b | 2.24 ± 0.05 ^a |
| C18:1 (n-9) | 19.3 ± 0.2 ^a | 20.3 ± 0.1 ^b |
| C20:1 (n-9) | 0.68 ± 0.05 ^a | 0.74 ± 0.07 ^b |
| Polyunsaturated | | |
| C18:2 (n-6) | 26.6 ± 0.2 ^a | 28.0 ± 0.1 ^b |
| C18:3 (n-3) | 5.3 ± 0.2 ^a | 5.5 ± 0.1 ^a |
| C20:2 (n-6) | 0.8 ± 0.1 ^a | 0.8 ± 0.1 ^b |
| Total SFA | 41.1 ± 0.4 ^a | 42.6 ± 0.3 ^b |
| Total MUFA | 26.3 ± 0.4 ^b | 23.3 ± 0.3 ^a |
| Total PUFA | 32.7 ± 0.4 ^a | 34.4 ± 0.3 ^b |
| PUFA/SFA | 0.79 ± 0.02 ^a | 0.81 ± 0.01 ^b |
| Tocopherols (mg/100 g fw) | | |
| α-Tocopherol | 0.060 ± 0.001 ^a | 0.05 ± 0.01 ^a |
| γ-Tocopherol | 0.030 ± 0.002 ^a | 0.03 ± 0.01 ^a |
| δ-Tocopherol | 0.010 ± 0.001 | nd |
| Total tocopherols | 0.10 ± 0.01 ^a | 0.07 ± 0.001 ^a |

In each row different letters mean significant differences ($p < 0.05$). C6:0 Pentanoic acid, C12:0 Lauric acid, C14:0 myristic acid, C15:0 pentadecanoic acid, C16:0 palmitic acid, C16:1 palmitoleic acid, C17:0 heptadecanoic acid, C18:0 stearic acid, C18:1n9c oleic acid, C18:2n6c linoleic acid, C18:3n6 γ-linolenic acid, C18:3n3 α-linolenic acid, C20:0 arachidic acid, C20:1 eicosenoic acid, C20:2 eicosadienoic acid, C22:0 behenic acid, C23:0 tricosylic acid, C24:0 lignoceric acid. SFA: saturated fatty acids; MUFA: monounsaturated fatty acids; PUFA: polyunsaturated fatty acids.

2014). To the best of our knowledge, the present study is the first report on the organic acid composition of *A. stipularis* roots and rhizomes, so there are no data to compare them with the present results. However, previous studies reported the organic acid composition of other *Asparagus* species (Jime'nez-Sánchez et al., 2016; Slatnar et al., 2018). Zhang et al. (2020) conducted a study on both green and white *Asparagus*. Results showed the presence of several organic acids such as fumaric, ascorbic, malic and citric acids.

The oxalic acid/Ca ratio was 0.752 and 0.568 for roots and rhizomes, respectively, which is much lower than 2.5, which suggests that the consumption of *A. stipularis* samples doesn't reduce the bioavailability of dietary Ca by the formation of insoluble calcium oxalate (Adouni et al., 2018).

The fatty acid profiles of the roots and rhizomes of *A. stipularis* were determined (Table 3). Saturated fatty acids (SFA) predominated over monounsaturated fatty acids (MUFA) and polyunsaturated fatty acids (PUFA) in both samples, though presenting the highest percentage in rhizomes ($42.6 \pm 0.3\%$). Among them, palmitic acid (C16:0) presented the main contribution ($24.10 \pm 0.07\%$). The roots showed higher levels of MUFA ($26.3 \pm 0.4\%$) than rhizomes ($23.3 \pm 0.3\%$), with prevalence of oleic acid (C18:1) in both samples. On the other hand, PUFA were higher in rhizomes ($34.4 \pm 0.3\%$), mainly due to linoleic acid ($28.0 \pm 0.1\%$). The ratio of PUFA/SFA was higher than 0.45 in both samples, which is considered an adequate nutritional relation for food products (Simopoulos, 1997). Very small contents of tocopherols were determined in roots and rhizomes of *A. stipularis*, with α and γ isoforms present in both samples, and δ -tocopherol present only in the roots (Table 3).

Besides their sensory properties, volatile aroma compounds present in plant foods demonstrated important biological functions for human health as well as for plant and food protection (Hubert et al., 2008; Boulogne et al., 2012; Keiler et al., 2013). Thus, volatile aroma can be considered as potentially safe alternatives to chemical flavoring in foods and functional products. Alcohols, esters, aldehydes, hydrocarbons, terpenoids, apocarotenes and other compounds were identified in the volatile profile of *A. stipularis* roots and rhizomes, being aldehydes and hydrocarbons the main chemical classes (Table 4). Limonene and nonanal were the major volatile compounds detected, presenting the highest level in roots ($20.8 \pm 0.1\%$ and $12.6 \pm 0.1\%$, respectively). Limonene is added as a flavoring compound in perfumes, soaps, and foods (Whysner & Williams, 1996), while nonanal has been used for the improvement of oral breath odor and the diagnosis of some human diseases (Itoh et al., 2013).

3.2. Bioactive compounds in *A. stipularis* infusions

3.2.1. Organic acids

The organic acid profile of *A. stipularis* infusions is given in Table 5. Oxalic and citric acids were the major acids detected in the preparation of the roots, being oxalic acid the most abundant one (17.60 ± 0.05 mg/g extract). Only citric acid was present in the rhizome infusion, and its content was lower than the one determined in that of the roots. Recent studies have shown that organic acids exert some health benefits such as antioxidant (Liu et al., 2019) and antimicrobial activities (Mani-Lopez et al., 2012).

3.2.2. Phenolic compounds

Asparagus can be considered as products of interest due to their contents in phytochemicals, such as phenolic compounds (Fuente-s-Alventosa et al., 2013). As a first approach, total phenolic (TPC) and flavonoids contents (TFC) of roots and rhizomes infusions were estimated by spectrophotometric methods (Table 5). The roots extract was richer in phenolic compounds than rhizomes, where no flavonoids were detected in the TFC assay. Similarly, Hamdi et al. (2016) detected flavonoids in ethanol extracts of leaves and pericarps of *A. albus*, but not in those of rhizomes. It must be said, however, that it was not possible to

Table 4

Volatile compounds characterized in *Asparagus stipularis* roots and rhizomes.

| Compound | L.r.i.* | Content (%) | |
|---|---------|-------------------------|--------------------------|
| | | Roots | Rhizomes |
| Hexanal | 802 | 1.3 ± 0.1 ^a | 8.5 ± 0.1 ^b |
| 1-Hexanol | 869 | nd | 0.9 ± 0.1 |
| Heptanal | 901 | nd | 1.2 ± 0.1 |
| Benzaldehyde | 962 | 1.7 ± 0.1 | nd |
| Hexanoic acid | 987 | 7.3 ± 0.2 ^a | 10.2 ± 0.1 ^b |
| Octanal | 1002 | 1.6 ± 0.1 ^b | 1.1 ± 0.1 ^a |
| <i>p</i> -Cymene | 1028 | nd | 1.0 ± 0.1 |
| Limonene | 1032 | 20.8 ± 0.1 ^b | 18.8 ± 0.2 ^a |
| γ -Terpinene | 1063 | 2.5 ± 0.1 ^a | 2.9 ± 0.1 ^b |
| 3-Octen-2-one | 1043 | nd | 2.9 ± 0.1 |
| Nonanal | 1104 | 12.6 ± 0.1 ^b | 12.2 ± 0.1 ^a |
| (<i>E,Z</i>)-3,5-Octadien-2-one | 1072 | nd | 2.6 ± 0.3 |
| (<i>E</i>)-2-Undecene | 1106 | 4.8 ± 0.1 ^b | 2.9 ± 0.2 ^a |
| Pentylisovalerate | 1108 | 1.7 ± 0.1 | nd |
| (<i>Z</i>)-3-Hexenyl isobutyrate | 1143 | 2.3 ± 0.1 ^b | 1.7 ± 0.2 ^a |
| 2-Ethylhexyl acetate | 1155 | 9.9 ± 0.1 ^b | 3.8 ± 0.3 ^a |
| <i>trans</i> -Linalool oxide (furanoid) | 1090 | nd | 1.4 ± 0.2 |
| (<i>E,E</i>)-3,5-octadien-2-one | 1093 | nd | 1.2 ± 0.1 |
| α -Terpineol | 1191 | 1.8 ± 0.1 | nd |
| Linalool | 1101 | nd | 1.3 ± 0.2 |
| Safranal | 1197 | 3.4 ± 0.3 | nd |
| Decanal | 1206 | 5.7 ± 0.1 ^b | 5.3 ± 0.2 ^a |
| <i>n</i> -Tridecane | 1300 | 1.9 ± 0.1 | nd |
| (<i>E</i>)-2-Nonenal | 1163 | nd | 0.9 ± 0.1 |
| <i>n</i> -Undecane | 1100 | nd | 1.2 ± 0.1 |
| <i>n</i> -Dodecane | 1200 | nd | 1.00 ± 0.07 |
| <i>n</i> -Tridecane | 1300 | nd | 1.0 ± 0.1 |
| α -Terpinyl acetate | 1352 | 1.3 ± 0.2 ^a | 1.6 ± 0.2 ^b |
| 2-Methylbutyl heptanoate | 1355 | 2.1 ± 0.2 | nd |
| 2-Methylundecanal | 1368 | 1.3 ± 0.1 | nd |
| α -Copaene | 1377 | 3.6 ± 0.1 ^b | 2.1 ± 0.1 ^a |
| <i>n</i> -Tetradecane | 1400 | 3.2 ± 0.1 ^b | 1.1 ± 0.1 ^a |
| β -Caryophyllene | 1419 | 2.2 ± 0.2 | nd |
| β -Bisabolene | 1508 | nd | 3.4 ± 0.2 |
| <i>trans</i> - α -Bergamotene | 1437 | nd | 2.7 ± 0.2 |
| (<i>E</i>)-Geranylacetone | 1455 | 2.2 ± 0.1 ^b | 1.9 ± 0.1 ^a |
| <i>n</i> -Pentadecane | 1500 | 2.4 ± 0.1 | nd |
| Monoterpene hydrocarbons | – | 23.6 ± 0.4 ^b | 22.6 ± 0.3 ^a |
| Oxygenated monoterpenes | – | 3.2 ± 0.1 ^a | 4.3 ± 0.2 ^b |
| Sesquiterpene hydrocarbons | – | 5.7 ± 0.1 ^a | 8.2 ± 0.2 ^b |
| Apocarotenes | – | 5.6 ± 0.3 ^b | 1.70 ± 0.07 ^a |
| Non-terpene derivatives | – | 58.8 ± 0.2 ^a | 58.4 ± 0.2 ^a |
| Total identified | | 97.80 | 95.80 |

*L.r.i.: linear retention index; In each row different letters mean significant differences ($p < 0.05$); nd: not detected.

Table 5

Organic acids and phenolic fractions in infusion extracts of *Asparagus stipularis* roots and rhizomes.

| | Roots infusion | Rhizomes infusion |
|---|---------------------------------|--------------------------------|
| Organic acids (mg/g extract) | | |
| Oxalic acid | 17.60 ± 0.05 | tr |
| Malic acid | tr | tr |
| Citric acid | 7.6 ± 0.3 ^b | 1.31 ± 0.03 ^a |
| Total organic acids | 25.2 ± 0.2^b | 1.31 ± 0.03^a |
| Total polyphenols (mg GAE/g extract) | 58.40 ± 2.10^b | 22.0 ± 0.70^a |
| Total flavonoids (mg CE/g extract) | 20.90 ± 0.30 | nd |

In each row different letters mean significant differences ($p < 0.05$); tr: traces; nd: not detected.

find additional flavonoids in the HPLC analysis of single phenolic rhizomes or roots.

According to Petropoulos et al. (2017), the phenolic profiles varies depending on plant part, genotype, and growing and processing conditions, leading to large variations in both content and composition. Phenolic compounds are plant secondary metabolites involved in the

Table 6

Retention time (Rt), wavelengths of maximum absorption in the UV-vis region, mass spectrometric data, tentative identification and quantification of individual phenolic compounds in *Asparagus stipularis* rhizomes and roots infusions (mg per 100 g of dw).

| Peak | Rt (min) | λ_{max} (nm) | Pseudomolecular ion $[M-H]^-$ (m/z) | MS^2 (m/z) | Tentative identification | Quantification (mg/100 g dw) | |
|---------------------------------|----------|----------------------|---|-------------------------|-------------------------------|--------------------------------|---------------------------------|
| | | | | | | Rhizomes | Roots |
| 1 | 9 | 294, 323 | 179 | 135 | Caffeic acid | 0.16 ± 0.03 ^a | 19.83 ± 0.1 ^b |
| 2 | 10.2 | – | 353 | 179, 135 | Caffeolquinic acid | nd | 3.25 ± 0.02 |
| 3 | 15 | 323 | 267 | 252, 193, 175, 160, 133 | Feruloyl glycerol | 1.69 ± 0.13 ^a | 5.02 ± 0.3 ^b |
| 4 | 29.2 | 315 | 413 | 134, 160, 193, 145, 119 | Coumaroylferuloyl glycerol I | 1.11 ± 0.02 ^a | 12.48 ± 0.03 ^b |
| 5 | 29.7 | 325 | 443 | 193, 175, 160, 134 | Diferuloyl glycerol I | 4.79 ± 0.03 ^a | 16.32 ± 0.64 ^b |
| 6 | 30.5 | 312 | 413 | 134, 160, 193, 145, 119 | Coumaroylferuloyl glycerol II | 0.71 ± 0.05 | nd |
| 7 | 30.9 | 324 | 443 | 193, 175, 160, 134 | Diferuloyl glycerol II | 0.077 ± 0.01 | nd |
| Total phenolic compounds | | | | | | 8.56 ± 0.13^a | 56.93 ± 0.70^b |

In each row different letters mean significant differences ($p < 0.05$); nd: not detected.

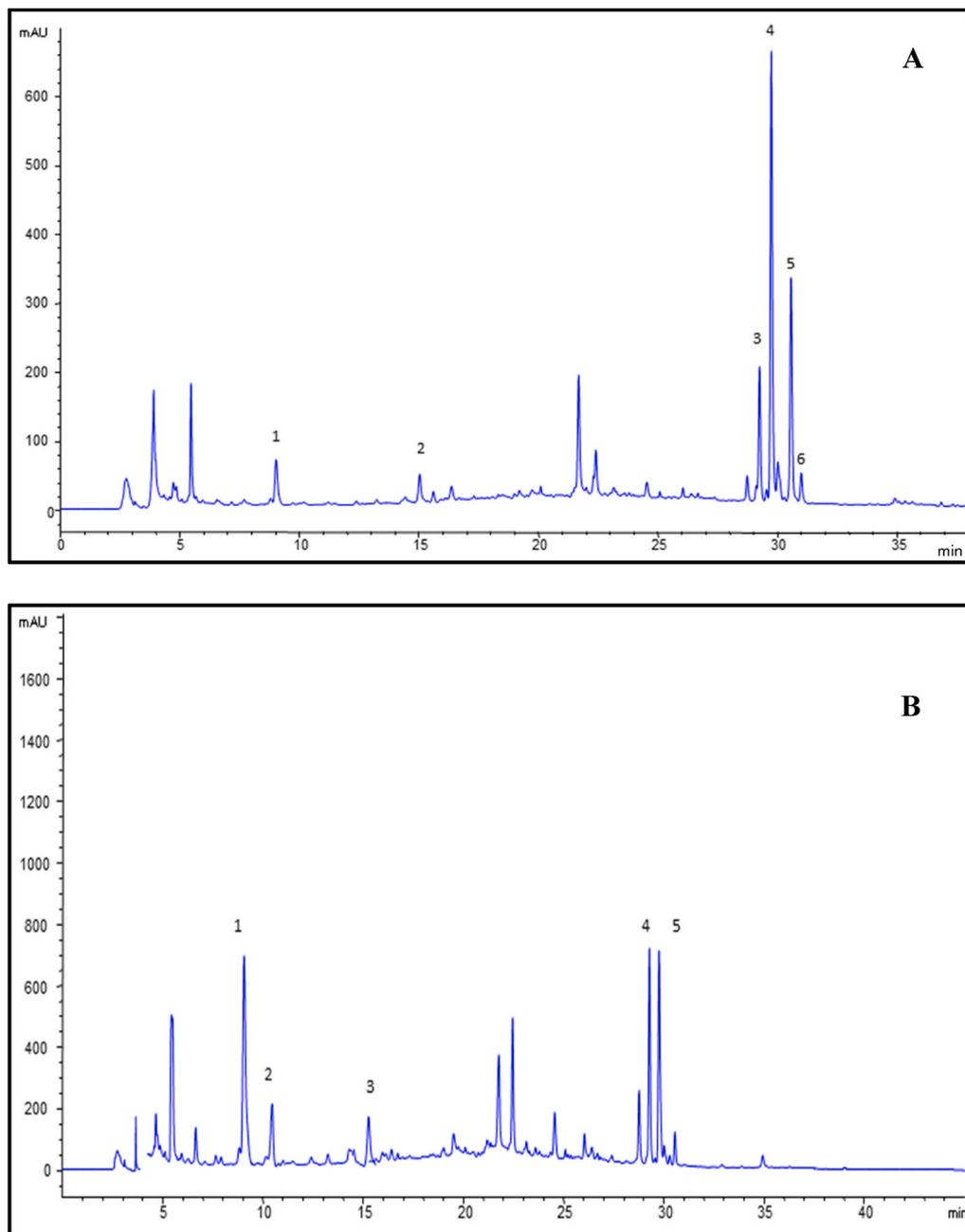


Fig. 1. HPLC-UV (330 nm) chromatogram of the infusions from rhizomes (A) and roots (B) from *Asparagus stipularis*.

natural mechanisms of defence against biotic and abiotic stresses, and with relevant biological activities that have been related to human health protection (Quideau et al., 2011). Phenolic compounds have been reported to exhibit antioxidant, inflammatory, antimicrobial, and anti-proliferative activities (Ferreira et al., 2017; Santos-Buelga et al., 2019).

HPLC-DAD/ESI-MS analysis was performed to obtain more insights on the phenolic composition of *A. stipularis* extracts. Peak characteristics, tentative identifications and quantification are presented in Table 6, and exemplificative chromatograms for the extracts of roots and rhizomes are shown in Fig. 1. Seven hydroxycinnamoyl derivatives were detected in both samples. Peak 1 was positively identified as caffeic acid by comparison with a commercial standard, while the remaining compounds were tentatively identified based on their UV and mass spectra characteristics. Compound 2 ($[M-H]^-$ at m/z 353) corresponded to a caffeoylquinic acid as shown by the product ions at m/z 179 (caffeic acid-H) and 135 (caffeic acid-CO₂-H). The other five compounds were associated to ferulic acid derivatives, supported by the presence of the ion at m/z 193 (ferulic acid-H), the presence of cumaroyl moieties in peaks 4 and 6 was confirmed by the ion at m/z 145 (coumaric acid-H₂O-H) and the decrease in the maximum UV wavelength approaching to that of coumaric acid (310 nm). Thus, the following identities were assigned as feruloyl glycerol (compound 3), two diferuloyl glycerol isomers (compounds 5 and 7) and two coumaroyl-feruloyl glycerol isomers (compounds 4 and 6). These compounds and similar MS/MS fragmentation patterns were also reported in the infusion of *A. stipularis* spears previously published (Adouni et al., 2018), and described in green *Asparagus* (*Asparagus officinalis*) samples by Jimenez-Sanchez et al. (2016). The phenolic composition profile of both analysed extracts was relatively similar, nevertheless, but with notable quantitative differences (Table 6), with phenolic contents that were approximately seven times higher in roots than the rhizomes (56.93 ± 0.70 vs 8.56 ± 0.13 mg/100 g dw).

3.3. Bioactivity of *A. stipularis* extracts

Antioxidant activities should be measured using different methods, with various concepts, applications, mechanisms of action and ways of expressing results (Oliveira et al., 2009). For this reason, the antioxidant activity of *A. stipularis* infusions was determined by DPPH, ABTS scavenging and FRAP assays, and the results were expressed as EC₅₀ values (Table 7). As observed, the roots infusion revealed much higher antioxidant activity in all the different tests than that of the rhizomes (EC₅₀ values of 0.44 ± 0.05 vs 4.70 ± 0.10 , for DPPH; 0.98 ± 0.03 vs 7.38 ± 0.07 , for ABTS and 0.64 ± 0.05 vs 7.80 ± 0.20 , for FRAP). This greater activity could be related to the higher TPC and TFC levels (Table 5) and phenolic contents (Table 6) in the root infusion. Positive correlations between the antioxidant activity of *Asparagus* species and TFC have been already described in previous studies (Hamdi et al., 2016; Sun et al., 2007). Indeed, it was reported that in green *Asparagus*, higher phenolic contents were associated to a higher antioxidant capacity (Rodríguez et al., 2005). Furthermore, the greater content in organic acids found in root infusion could also contribute to its higher antioxidant effectiveness compared to rhizomes. The antioxidant potential of distinct organic

Table 7
Antioxidant activity and cytotoxicity of *Asparagus stipularis* infusions.

| | Roots Infusion | Rhizomes Infusion |
|--|--------------------|----------------------|
| Antioxidant activity (EC ₅₀ , mg/mL) | | |
| DPPH scavenging activity | 0.44 ± 0.005^a | 4.70 ± 0.10^b |
| ABTS scavenging activity | 0.98 ± 0.03^a | 7.38 ± 0.07^b |
| Reducing power assay | 0.64 ± 0.01^a | 7.80 ± 0.20^b |
| Cytotoxicity (GI₅₀ value, µg/mL) | | |
| PLP2 | >400 | >400 |

In each row different letters mean significant differences ($p < 0.05$).

acids, for instance, oxalic, malic, citric or succinic acids, has been highlighted by different authors and based on their metal chelating ability (Seabra et al., 2006). Thus, the contribution of organic acids to the antioxidant activity of *A. stipularis* samples cannot be overlooked.

The antioxidant properties of some *Asparagus* species have been previously demonstrated (Lee et al., 2014; Rodríguez et al., 2005). Nevertheless, the outcomes cannot be compared to those obtained herein due to the differences in the types of assays, extracts and sample preparation and expression of the results.

Regarding cytotoxicity, none of the preparations showed a toxic effect in the porcine liver primary cell culture (GI₅₀ > 400 µg/mL), suggesting that they can be used as safe alternative antioxidant ingredients for medicine or pharmaceutical and food applications.

3.4. Encapsulation of *A. stipularis* root extracts into PLGA nanoparticles

Plant extracts are complex mixtures of chemicals. Yet, their application has been sometimes restricted because of their instability during pharmaceutical and food processing, storage (pH, oxygen, temperature, light) as well as digestion (pH, enzymes) (Fang & Bhandari, 2010). Research has been more usually centred on the composition of plant extracts, while solutions that permit their safe, efficient, and direct applications are scarcer. One of the recent ways to overcome some of the outlined limitations is the encapsulation of bioactive compounds into polymeric nanoparticles, to provide protection and allow their delivery in a sustained, controlled or targeted manner (Armendáriz-Barragán et al., 2016). Thus, in the current study, a formulation of PLGA nanoparticles encapsulating the extract of *A. stipularis* root infusion, the most potent of the analysed ones, was developed for potential application in foods, drugs or cosmetics.

3.4.1. Physicochemical properties of nanoparticles (NPs)

The particle size, PDI and ZP of the unloaded and loaded prepared NPs are shown in Table 8. Unloaded NPs were produced as controls and their particle size was 243 ± 4 nm, the PDI around 0.1 and the ZP negative (-21 mV), which are suitable characteristics for compound administration by different paths. The *Asparagus*-loaded NPs have a slightly higher particle size (261 ± 5 nm) than the unloaded ones, which is indicative of the incorporation of the extract into the NPs. Silva et al. (2014) reported particle size values for PLGA 50:50 NPs loaded with phenolic compounds between 140 and 250 nm, whereas Pereira et al. (2015) obtained particle sizes of 300 nm during the encapsulation of methanol extracts of *Lavandula pedunculata* and *Lavandula stoechas* in PLGA NPs. Tachaprutinun et al. (2014) illustrated that NPs with small particle sizes can be suitable pharmaceutical carriers for extracts intended for skin application, as they can easily penetrate intra- and intercellular spaces and through the hair follicles. Pray and Yakine (2009) reported that, when applied with antimicrobial functionality, those NPs do not affect the appearance or texture of food products, as they are invisible to the human eye and cannot be detected in the mouth. Concerning the PDI, which is a dimensionless measure of the broadness of particle size distribution, the results were close to 0.1 for both unloaded and loaded NPs. This result shows a mono- and homogenous dispersity and, consequently, a better particle size distribution (Zigoneanu et al., 2008). As for the ZP, which is the surface charge of the nanoparticles, a value of -35.9 ± 0.5 mV was found for loaded NPs. The negative ZP value was due to the presence of terminal carboxylic groups

Table 8

Average particle size, polydispersity index (PDI), zeta potential (ZP) and encapsulation efficiency (EE) of PLGA nanoparticles (Np) unloaded and loaded with an *Asparagus stipularis* root infusion extract.

| | Particle size (nm) | PDI | ZP (mV) | EE (%) |
|-------------------------|--------------------|-----------------|-----------------|------------|
| Unloaded PLGA Np | 243 ± 4 | 0.11 ± 0.02 | -21 ± 3 | - |
| Loaded PLGA Np | 261 ± 5 | 0.13 ± 0.02 | -35.9 ± 0.5 | 83 ± 3 |

in the polymers (Jahan et al., 2015). Klang et al. (2010) suggested ZP of ± 30 mV as a standard value for stable nanoemulsions, which ensures a high-energy barrier that stabilizes emulsions. Hence the formulations prepared herein are physically stable with little tendency to particle aggregation.

3.4.2. Encapsulation efficiency (EE)

The EE values differ considerably from study to study and many factors influenced them, such as the encapsulated material, the polymer molecular weight as well as the lactide:glycolide ratio (Wischke et al., 2008). In the current study, the EE of *Asparagus* roots infusion was 82.87% (Table 8). This value presents a good achievement for plant extracts encapsulated into PLGA in comparison with the literature. The EE value is higher than the previously reported values for PLGA nanoparticles containing plant extracts. Silva et al. (2014) described EE of 1.15% and 10.61% for hydroalcoholic extracts of guava and passion fruit encapsulated in PLGA 65:35 NPs and PLGA 50:50 NPs, respectively. EE of 39% in PLGA 65:35 NPs and of 48% in PLGA 50:50 NPs were described for cinnamon roots extract by Hill et al. (2013).

3.4.3. Scanning electron microscopy (SEM)

Both formulations (unloaded and loaded NPs) were analysed by SEM after freeze-drying with and without trehalose, a classical cryoprotectant. The SEM analysis showed NPs with a spherical morphology and a smooth surface (Fig. 2). Results are in agreement with literature about PLGA NPs (Fonte et al., 2014, 2015). Additionally, the similarity in particle size seen in the SEM analysis was also harmonious, with the physicochemical properties reported in Table 8. As expected, the NPs freeze-dried in the presence of trehalose appear to have a coating that makes them closer, as a slight aggregation, due to the positive cryoprotectant effect of this substance (Fonte et al., 2015).

3.4.4. Fourier transform infrared (FTIR) spectroscopy analysis

FTIR spectra of *Asparagus* roots infusion, physical mixture, PLGA and unloaded nanoparticles (used as controls) and the *Asparagus*-loaded NPs were performed after freeze-drying and shown in Fig. 3. The PLGA spectrum revealed the appearance of the characteristic peak between 1750 and 1760 cm^{-1} , which matching the C=O stretching, and the presence of the peak at 3000 cm^{-1} , corresponding to the C-H stretching (Singh et al., 2014). The comparison between the PLGA spectrum and the physical mixture and the NPs spectra indicates a slight deviation to the right at the C-H stretching of the PLGA, probably due to the interaction between the PLGA and the other substances. Additionally, the *Asparagus*-loaded NPs showed not only that deviation but also a rise in

the intensity of the same peak, which can be a result of the superimposition of the extract peak at the same wavenumber, as it could be observed in the *Asparagus* (roots infusion) spectra. It was also important to mention that the *Asparagus* spectra reveals a peak between 1650 and 1600 cm^{-1} corresponding to the C=C stretching. This peak is also observed in the physical mixture; however, it is not present in the *Asparagus*-loaded NPs, which may corroborate the efficient encapsulation of the extract.

3.4.5. Differential scanning calorimetry (DSC) analysis

The loaded NPs were also evaluated by DSC, as well as the controls (*Asparagus* roots infusion, PLGA, physical mixture and unloaded NPs), after freeze-drying, and the results are depicted in Fig. 4. The thermograms have a similar melting profile among all samples. Furthermore, the characteristic endothermic peak of PLGA at 48 °C (Erбетта et al., 2012) was observed in the control samples and in the loaded NPs. Additionally, the *Asparagus* roots infusion thermogram displayed an exothermic peak at 150 °C, which was not observed in the *Asparagus*-loaded NPs, probably due to the encapsulation of the extract into the nanoparticles, that stabilizes the systems. This observation also supports the EE results and the physicochemical properties obtained for the nanocarrier.

4. Conclusions

The present work provides new information on the nutrient and phytochemical composition of *A. stipularis* roots and rhizomes. The samples resulted to be a good source of essential and non-essential compounds, such as proteins, lipids, minerals, organic acids and phenolic acids. The evaluation of the antioxidant potential using various *in vitro* methods demonstrated that the infusion of the roots possessed relevant antioxidant capacity. None of the studied samples was toxic to normal cells. It was also demonstrated that the root infusion extract can be successfully loaded into PLGA nanoparticles. The obtained NPs showed a spherical morphology and high encapsulation efficiency, and the successful loading of the root extract within the developed NPs was also confirmed by FTIR and DSC analysis. The application of nanotechnology could open new perspectives for the potential use of *A. stipularis* extracts as innovative nutraceuticals. Such nanoparticles produced using plants have been used in various applications for human benefit. Thus, the developed nanoparticles are expected to be used as delivery systems for bioactive compounds of *A. stipularis* and they could be used as an innovative dietary supplement for pharmaceutical and food industry.

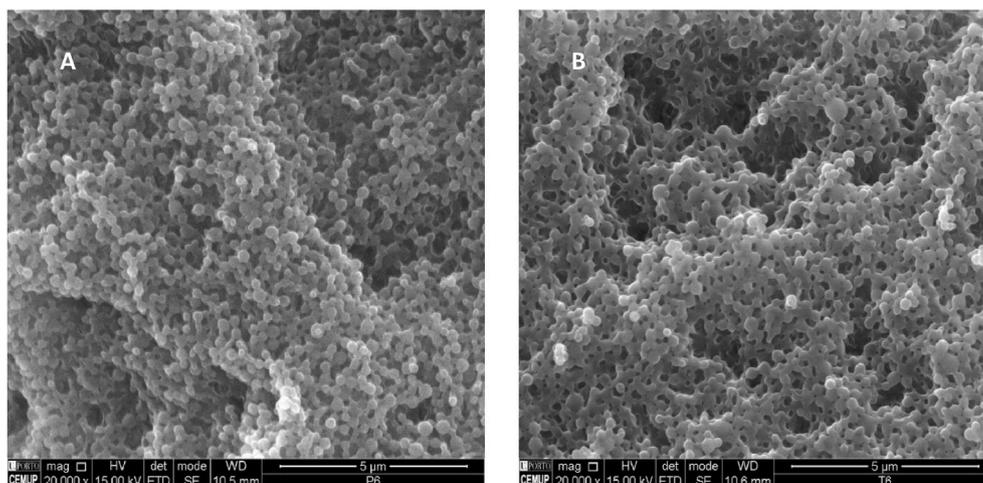


Fig. 2. SEM microphotographs of *Asparagus* loaded PLGA nanoparticles, after freeze-drying without (A) and with (B) 5% (w/v) of trehalose, at magnification of 20,000 \times . The scale bar of the microphotographs at the bottom right of the images corresponds to 5 μm .

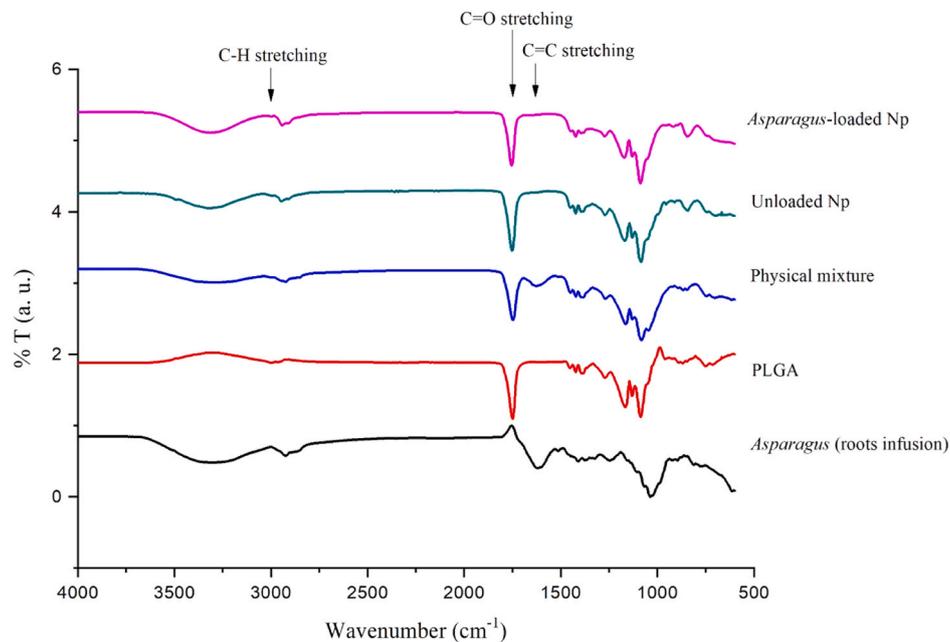


Fig. 3. FTIR spectra of *Asparagus* (roots infusion), PLGA, physical mixture and unloaded nanoparticles (NP), as controls, and *Asparagus*-loaded NP after freeze-drying.

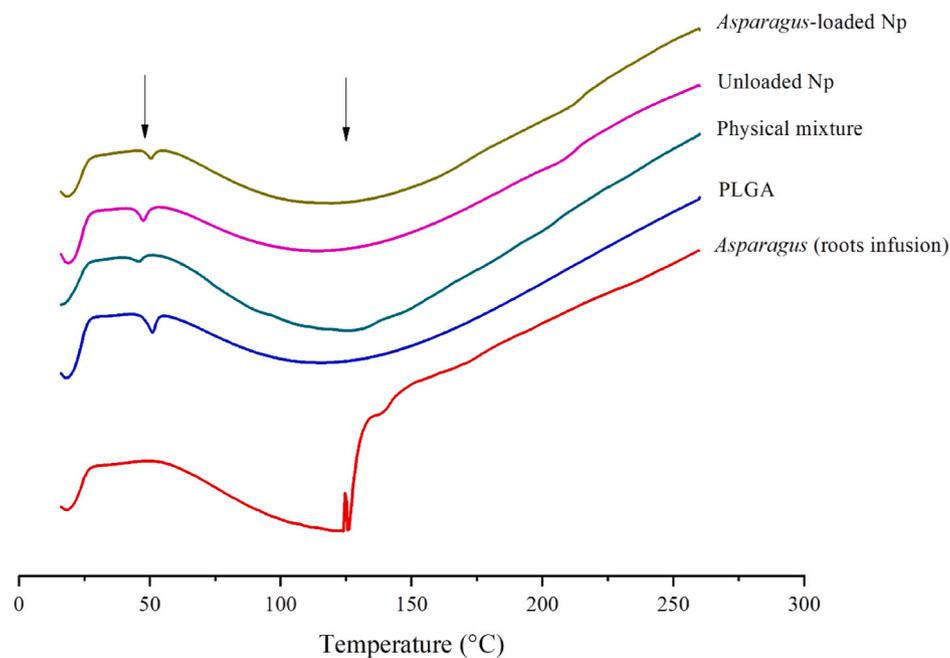


Fig. 4. DSC thermogram of *Asparagus* (roots infusion), PLGA, physical mixture and unloaded nanoparticles (NP), as controls, and *Asparagus*-loaded NP after freeze-drying.

Authors' contributions

Khaoula Adouni was responsible for the collection and preparation of plant materials, performed the experiments, data analysis, and drafted the manuscript; **Guido Flamini** performed the experiments related to volatile compounds; **Ana M. González-Paramás** contributed to the HPLC-MS analysis. **Ana Júlio** and **Patrícia Filipe** contributed to the FTIR and DSC results and writing the manuscript.

Isabel C.F.R. Ferreira and **Ângela Fernandes** contributed to the experiments related to tocopherols, organic acids and cytotoxicity; **Pedro Fonte**, **Lotfi Achour**, **Patrícia Morales**, **Virginia Fernández-Ruiz**, **Celestino Santos-Buelga**, **Patrícia Rijo**, **Sofia A. Costa Lima**

and **Salette Reis** organized this work and contributed to writing the manuscript. All authors read and approved the final manuscript.

Declaration of competing interest

The authors confirm that they have no conflicts of interest.

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