Chemical and physicochemical changes in *Serrana* goat cheese submitted to extra-long ripening periods

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**A R T I C L E I N F O**

Article history:
Received 27 June 2017
Received in revised form 18 August 2017
Accepted 26 August 2017
Available online 28 August 2017

Keywords:
Proximate composition
Color parameters
Fatty acids
Extra-long ripening
Linear discriminant analysis

**A B S T R A C T**

Goat milk cheese is highly appreciated since ancient times. However, besides being consumed close to its production date, according to the common practice, there is an increasing trend towards consuming goat milk cheese submitted to long ripening periods. Throughout this process, several chemical changes are expected, justifying the different properties presented by ripe cheese in comparison to fresh one. Accordingly, proximate composition, salt percentage, saponifiable matter, color parameters, pH and fatty acids were compared in cheese submitted to medium, long and extra-long ripening. The parameters showing significant changes among different ripening periods were effectively identified, highlighting fatty acids as the ones with highest variation. Actually, fatty acids profiles were successfully assayed as a chemical fingerprint of goat milk cheese submitted to different ripening periods. This result raised up the possibility of discriminating goat milk cheese ripening period using a single, fast and reliable chromatographic methodology.

**1. Introduction**

Cheese represents one of the most consumed milk derivatives, being highly appreciated worldwide, available in different forms and types and produced from different types of milk. Cheese prepared with sheep or goat milk, owing to its high quality (as detailed below) and the low volume of milk produced by these animals, is often considered as a specialty product, representing important products in Mediterranean economy (Bontinis, Mallatou, Pappa, Massouras, & Alichanidis, 2012; González-Martín, Palacios, Revilla, Vivar-Quintana, & Hernández-Hierro, 2017; Soto-Barajas et al., 2013). In Portugal, as in most countries, the productivity of goat milk is not comparable to that of cow milk; nevertheless, the production volume is considerable, reaching annual volumes around 30 million litres (INE, 2015). Furthermore, cheese produced from goat milk is greatly valued for its organoleptic characteristics, being also acknowledged for its higher digestibility, alkalinity and beneficial effects in human health (Park, 2007).

Traditionally, goat milk cheese was mainly consumed without being submitted to a ripening process, contrarily to what was practiced with cheese produced from sheep's milk. However, there is an increasing tendency to consume ripe cheese made from goat milk. As this ripening proceeds, different physical and chemical changes are expected to occur. In fact, cheese ripening is a complex procedure that is accompanied by pH variation, breakdown of proteins, slow accumulation of amino acids and lipid degradation (Farahani, Ezzatpanah, & Abbasi, 2014; González-Martín et al., 2017). Besides proteolysis and lipolysis, glycolysis is an equally important chemical process occurring in the course of cheese ripening (Delgado, González-Crespo, Cava, & Ramírez, 2012; Hayaloglu, Tolu, & Yasar, 2013).

Glycolysis involves several reactions, mainly associated to lactose degradation (Bezerra et al., 2017), while proteolysis is primarily conducted by enzymes naturally found in milk, coagulant agents and microbial enzymes produced by lactic acid bacteria (Bezerra et al., 2016). The overall process is influenced by several elements, namely pH, salt-in-moisture, salt content, humidity, storage temperature and time, enzymes present in milk, microorganisms and rennet (Hayaloglu et al., 2013; Park, 2001). Finally, lipolysis induces specific changes in fatty acids profile (Park, 2001; Poveda & Cabezas, 2006).

The occurrence of all these modifications is one of the main reasons leading to the exclusive characteristics in different cheese varieties (Farahani et al., 2014; González-Martín et al., 2017;
Therefore, it would very interesting to understand how chemical profile evolves throughout ripening, in order to identify the most significant changes characterizing each ripening period.

In this work, different extended ripening periods (up to a maximum of 24 months) were applied to highly appreciated Serrana goat-milk cheeses. Besides assessing physicochemical modifications in each assayed ripening period, the main objective was to evaluate if the profile changes were significant enough to distinguish each ripening period.

2. Materials and methods

2.1. Standards and reagents

The fatty acids methyl ester (FAME) reference standard mixture 37 (standard 47885-U) was purchased from Sigma (St. Louis, MO, USA), as also other individual fatty acid isomers. Methanol and all other chemicals were of analytical grade and obtained from common sources; nitrate sodium was purchased from Panreac S.A. (Barcelona, Spain) and potassium chromate from Merck KgaA (Darmstadt, Germany). Water was treated in a Mili-Q water purification system (TGI Pure Water Systems, USA).

2.2. Samples and cheese-making procedure

Goat-milk cheese was made in an industrial unit (LEICRAS, Cooperativa de Produtores de Leite de Cabra Serrana, Miranda, Northeast of Portugal), with raw milk from Serrana goat breed, salt and rennet from animal origin. After ripening, cheese maintained the traditional cylindrical shape. Its classification as Protected Designation of Origin requires the fulfilment of all rules stipulated in the product specification (including cabra Serrana's milk supply and milk production conditions, milk hygiene, manufacture of the product and labeling). The geographical area of production covers the municipalities of Alfandega da Fé, Alijó, Bragança (only 15 parishes), Carrazeda de Ansíes, Freixo de Espada à Cinta, Macedo de Cavaleiros, Miranda, Mogadouro, Murça, Torre de Moncorvo, Valpaços, Vila Flor and Vimioso.

Cheese samples, produced from the same milk lot, were divided into three lots: 3.032.2, 9.10.1 and 6.05.3. After 12 months of ripening three cheeses were separated from each lot and further submitted to the analyses described in section 2.3. After 6 additional months (18 months of ripening), three cheeses were separated and submitted to the same analyses. Finally, after 6 additional months (24 months of ripening) the remaining cheeses in each lot were also analyzed as described in section 2.3. Each ripening period was respectively designated as medium, long and extra-long ripening.

During the first month, samples were stored in a ripening chamber at 10–11 °C and relative humidity (RH) of 75–85%; in the second month, temperature varied around 9.5 and 11 °C and RH from 76 to 81%. After the second month, samples were stored in a preservation chamber at 10.5–12 °C and RH among 75–85%. Prior to analysis, all samples were lyophilized (FreeZone 4.5 model 7750031, Labconco, Kansas, USA), reduced to a fine dried powder (20 mesh) and mixed to obtain homogenized samples.

2.3. Quality analysis

2.3.1. Proximate composition

Cheese samples were analyzed for moisture, protein, fat and carbohydrate using the AOAC procedures (AOAC, 2016). Samples were lyophilized and reduced to a fine powder (20 mesh). Crude protein content (N × 6.28) was estimated by the macro-Kjeldahl method, using an automatic distillation and titration unit (model Pro-Nitro M Kjeldahl Steam Distillation System, Barcelona, Spain); crude fat was determined by extracting a known weight of powdered sample with petroleum ether, using a Soxhlet apparatus and total carbohydrates were calculated by difference according to the equation: 100 - (m_{moisture} + m_{proteins} + m_{fats}).

Results were expressed as g per 100 g of fresh weight (fw). Total energy was calculated according to the equation: 4 × (m_{proteins} + m_{carbohydrates}) + 9 × (m_{fats}) and expressed as kcal per 100 g of fresh weight (fw).

2.3.2. Salt determination

The NaCl concentration in cheese was determined according to Mohr’s method. Initially, 1 g of the sample was homogenized in 20 mL of distilled water and filtered through a Whatman No.4 paper. The powder was then homogenized with an additional portion of 20 mL and this procedure was repeated until a total of 5 time. Then the pH of the aqueous solution was adjusted to ≈8.5 with sodium hydroxide and 1 mL of potassium chromate solution (5%) was added. The mixture was titrated against AgNO3 (0.05 mol/L) until the appearance of the first reddish color (Ag_{2}CrO_{4} precipitate) (Ossai et al., 2014). The salt concentration was calculated using the following equation: Salt content % = \left[ \frac{\text{Vtitrated of AgNO}_3 \times 0.00292}{\text{[m} \text{sample}\times 100} \right] \times (1 mL of AgNO3 corresponds to 0.00292 g of NaCl). The results were expressed in g/100 g of fresh weight (fw).

2.3.3. Saponification

The saponification value was determined according to the method described in WHO (2016). The methodology started by dissolving around 2 g of neutral fat in 25 mL of an alcoholic solution of potassium hydroxide (0.5 mol/L). The mixture was then heated in a boiling plate for 20 min. Afterwards, 1 mL of phenolphthalein/ethanol (1%) was added and the excess of alkali was titrated with hydrochloric acid (0.5 mol/L). A blank test for titration volume was also performed with hydrochloric acid (0.5 mol/L). The normality 0.5 (N) and correction factor (F) 1.1098 was required. The saponification value was calculated from the following formula, and expressed in mg KOH: Saponification value = [(V_{required to titrate the blank} - V_{required to titrate the sample}) \times N \times F] = m_{mgKOH}/equivalent weight of KOH. The results were expressed in mg KOH per g of fresh weight (fw).

2.3.4. Color measurement

A Minolta spectrophotometer (model CR-400; Konica Minolta Sensing, Inc., Japan) was used to measure the color in three distinct zones of the cheese surface, daily, and these were considered as the average values. Using illuminant C and the diaphragm opening of 8 mm, the Hunter color L*, a* and b* values were reported through the computerized system using a color data software “Spectra Magic Nc” (version CM-S100W 2.03.0006, Konica Minolta company, Japan). The instrument was calibrated to standard white tiles before analysis (Spectra Magic NX Instruction Manual, Konica Minolta Sensing, Inc. (ver 2.0), 2009, Japan).

2.3.5. pH

5 g of cheese were macerated in 5 mL of distilled water, and the pH was measured using a calibrated digital pH meter (portable food and dairy pH meter HI 99161, Hanna Instruments, Woonsocket, RI, USA).

2.3.6. Fatty acids

Fatty acids were analyzed by gas chromatography using a flame ionization detector (GC-FID) and a capillary column (Pinela et al., 2016). Briefly, fatty acids present in fat extracted using a Soxhlet apparatus were methylated with methanol:sulphuric acid:toluene.
2:1:1 (v:v:v) during at least 12 h in a thermostat bath (50 °C, 160 rpm). Afterwards, deionized water was added to separate phases and fatty acid methyl esters (FAME) were extracted with diethyl ether by vortex shaking. The upper phase was dehydrated with anhydrous sodium sulfate and filtered (0.2 µm nylon filters) for chromatographic injections. The analysis was carried out in a DANI (model GC 1000) instrument equipped with a split/splitless injector, a FID detector at 260 °C and a Macherey-Nagel column (30 m × 0.32 mmID × 0.25 µm df). The oven temperature program was set as previously described (Pereira, Barros, Carvalho, & Ferreira, 2011). Fatty acids identification was made by comparing the retention times of FAME peaks from cheese samples with commercial standards. The results were recorded and processed using CSW 1.7 software (DataApex 1.7) and expressed in relative percentage.

2.4. Statistical analysis

Wherever possible, data were expressed as mean ± standard deviation. All statistical tests were performed at a 5% significance level using IBM SPSS Statistics for Windows, version 22.0. (IBM Corp., USA). A linear discriminant analysis (LDA) was used to evaluate the parameters with the most significant changes for each ripening period. The independent variables (physicochemical parameters) were included together and the assumptions of LDA were fully verified. To validate the significance of the canonical discriminating functions, Wilk's F test was used. A leaving-one-out cross validation procedure was carried out to assess the model performance.

3. Results and discussion

The main objective of this work was quantifying potential changes in the proximate composition and physicochemical properties of goat cheese submitted to largely extended ripening periods. Accordingly, the proximate composition, salt percentage, saponifiable matter, color parameters and pH values were evaluated, together with fatty acids profiles. In this first approach, changes induced by ripening stage were individually assessed in each cheese lot to prevent any biased effect induced by variability among lots. If statistically significant (p < 0.05), those changes were classified by Tukey’s HSD or Tamhane’s T2 tests for homoscedastic or heteroscedastic distributions, respectively.

3.1. Quality analysis

The mean values presented in Table 1 are given in g/100 g of fresh weight (fw). Moisture, fat and saponifiable matter were measured both in cheese rind (approximate thickness of 1 mm) and interior. Moisture percentage did not show (p > 0.05) high changes throughout maturation, especially in cheese interior, which maintained nearly the same values (11–14 g/100 g fw) along time. Regarding cheese rind, differences were a bit higher (15–20 g/100 g fw), with a general tendency to increase during ripening.

It could be expected that moisture tended to significantly decrease throughout cheese ripening. However, it should be reminded that most part of water content is lost on the first months of ripening, thereby being reasonable to expect low variations after the first 12 months. In fact, a similar increase in moisture content was previously verified in ripe cheese (Acerbi, Guiard, Guillaume, & Gontard, 2016; Kondyli, Pappa, & Svarnas, 2016). Furthermore, cheeses were maintained in chambers provided with constant RH (75–85%), which also contributed to maintain moisture values in cheese throughout long and extra-long ripening periods. Furthermore, during ripening process, cheeses are individually wiped with a moistened cloth. The same reasoning might be used to explain the higher moisture contents in cheese rind.

Conversely, fat contents tended to decrease, especially in cheese interior, as highlighted by the statistical differences measured in cheese samples from all lots (=45 g/100 g fw after 12 months to 40 g/100 g fw after 24 months). The reduction in fat content could, at first glance, seem surprising. However, this is a common process during cheese ripening, mostly due to lipolysis occurrence (Kondyli et al., 2016).

Proteins, which, together with lipids, represent the main components in cheese, were quantified in higher levels in cheese stored for longer periods, varying from 28 to 37 g/100 g fw. This increase might be considered as slightly surprising, since proteolysis processes are common during cheese ripening. Nevertheless, it must be bear in mind that this proteolysis occurrence tend to occur in the initial ripening months, and the values presented for medium ripe cheese samples already stored for at least 6 months, thereby beyond the period of expected significant proteolysis (Kondyli et al., 2016).

Contents in carbohydrates tended to decrease along ripening (14 g/100 g fw in medium ripe cheese to 11 g/100 g fw in extra-long ripe cheese). All together, these changes in macronutrients led to slighter lower energy values in extra-long ripe cheese (≈550 kcal/100 g fw, in comparison to ≈580 kcal/100 g fw in samples from medium ripe cheeses).

Salt contents were lower in extra-long ripe cheese, showing values around 1.7 g/100 g fw, which represent a significant decrease when compared to values quantified in cheese samples stored during 12 months (=2.5 g/100 g fw). The decrease in salt content is also unexpected, but it might be explained in part by the fact of having been quantified in fw basis, recalling that moisture level increased along ripening.

Finally, saponifiable matter levels showed a marked rise throughout ripening, increasing nearly 40%, from 39 to 42 mg KOH/g fw in medium-ripe cheese to 55–60 mg KOH/g fw in extra-long ripe cheese. The considerable increase in saponifiable matter is probably explained by the reduction in fat content, which might have been degraded to release saponifiable compounds.

The results for Hunter’s color L* (lightness), a* (redness) and b* (yellowness) parameters are presented in Table 2. The ripening period did not cause an unequivocal effect over the luminosity of cheese rind and interior, since different effects were obtained for each cheese lot. Even so, the interior of cheeses submitted to extra-long ripening showed a tendency to decreased L* values, while rind presented the opposite effect. In general, cheese samples exhibited high luminosity, as indicated by L* close to the positive limit (100) of the scale. The high values obtained for L* parameter are obviously explained by the glowing effect caused by the superficial fatty layer of cheese.

Regarding a* and b* values, the ripening process induced a clear increase in both parameters, independently of cheese lot. Values for a* parameter close to zero indicate that the studied samples presented negligible greenness (negative part of scale) or redness (positive part of scale). As it could be anticipated, b* values were positive (the negative part of this scale is associated with blue color, while the positive represents the yellow color), but not as high as it could be expected; this parameter might vary among −127 and +128, but the values obtained for cheese interior and rind were much lower (usually below 30). Even so, b* values tended to increase throughout ripening, most likely due to a slight browning effect, which is typical in cheese maturation.

Regarding pH (Table 2), the general tendency pointed towards a high stability around slightly acidic values. Changes observed among samples ripe for medium, long and extra-long periods represent only small fluctuations. The tendency to slight variations...
Table 1
Proximate composition (g/100 g fresh weight), energy (kcal/100 g fresh weight), salt (g NaCl/100 g fresh weight) and saponifiable matter (mg KOH/g fw). The results are presented as mean ± SD. Statistical classification was performed in each lot individually.a

<table>
<thead>
<tr>
<th>Lot</th>
<th>Maturation</th>
<th>Moisture</th>
<th>Fat</th>
<th>Protein</th>
<th>Carbohydrates</th>
<th>Energy</th>
<th>Salt</th>
<th>Saponifiable matter</th>
</tr>
</thead>
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<tr>
<td></td>
<td></td>
<td>interior</td>
<td>interior</td>
<td>interior</td>
<td>interior</td>
<td>interior</td>
<td>interior</td>
<td>interior</td>
</tr>
<tr>
<td>3.032.</td>
<td>Medium</td>
<td>13 ± 1 a</td>
<td>18 ± 1 c</td>
<td>45 ± 1 a</td>
<td>36 ± 1 b</td>
<td>28 ± 1 b</td>
<td>14 ± 2 a</td>
<td>575 ± 4 a</td>
</tr>
<tr>
<td></td>
<td>Long</td>
<td>12 ± 1 c</td>
<td>20 ± 1 b</td>
<td>42 ± 1 b</td>
<td>37 ± 1 a</td>
<td>35 ± 1 a</td>
<td>11 ± 2 b</td>
<td>565 ± 6 b</td>
</tr>
<tr>
<td></td>
<td>Extra-long</td>
<td>14 ± 1 a</td>
<td>22 ± 1 a</td>
<td>38 ± 1 c</td>
<td>34 ± 1 c</td>
<td>35 ± 1 a</td>
<td>13 ± 2 a</td>
<td>534 ± 5 c</td>
</tr>
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<td>Homoscedasticitya</td>
<td>p-value (n = 27)</td>
<td>&lt;0.001</td>
<td>&lt;0.001</td>
<td>&lt;0.001</td>
<td>&lt;0.001</td>
<td>&lt;0.001</td>
<td>&lt;0.001</td>
<td>&lt;0.001</td>
</tr>
<tr>
<td>One-way ANOVAa</td>
<td>p-value (n = 27)</td>
<td>&lt;0.001</td>
<td>&lt;0.001</td>
<td>&lt;0.001</td>
<td>&lt;0.001</td>
<td>&lt;0.001</td>
<td>&lt;0.001</td>
<td>&lt;0.001</td>
</tr>
<tr>
<td>9.101.</td>
<td>Medium</td>
<td>11 ± 1</td>
<td>18 ± 1 b</td>
<td>45 ± 1 a</td>
<td>36 ± 1 b</td>
<td>28 ± 1 c</td>
<td>15 ± 2 a</td>
<td>581 ± 3 a</td>
</tr>
<tr>
<td></td>
<td>Long</td>
<td>12 ± 1 a</td>
<td>19 ± 1 a</td>
<td>42 ± 1 b</td>
<td>38 ± 1 a</td>
<td>36 ± 1 b</td>
<td>11 ± 1 c</td>
<td>562 ± 5 b</td>
</tr>
<tr>
<td></td>
<td>Extra-long</td>
<td>12 ± 1 a</td>
<td>17 ± 1 b</td>
<td>40 ± 1 c</td>
<td>38 ± 1 a</td>
<td>37 ± 1 a</td>
<td>12 ± 1 b</td>
<td>550 ± 3 c</td>
</tr>
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<td>0.032</td>
<td>0.006</td>
<td>0.196</td>
<td>0.663</td>
<td>0.852</td>
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<td>0.110</td>
<td>&lt;0.001</td>
<td>&lt;0.001</td>
<td>&lt;0.001</td>
<td>&lt;0.001</td>
<td>&lt;0.001</td>
<td>&lt;0.001</td>
</tr>
<tr>
<td>6.053.</td>
<td>Medium</td>
<td>11 ± 1</td>
<td>16 ± 1 b</td>
<td>45 ± 1 a</td>
<td>36 ± 1 b</td>
<td>29 ± 1 c</td>
<td>14 ± 2 a</td>
<td>583 ± 10 a</td>
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<tr>
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<td>15 ± 1 a</td>
<td>42 ± 1 b</td>
<td>38 ± 1 a</td>
<td>35 ± 1 b</td>
<td>11 ± 1 b</td>
<td>568 ± 5 b</td>
</tr>
<tr>
<td></td>
<td>Extra-long</td>
<td>11 ± 1 a</td>
<td>15 ± 1 a</td>
<td>40 ± 1 c</td>
<td>36 ± 1 b</td>
<td>38 ± 1 a</td>
<td>11 ± 2 b</td>
<td>557 ± 7 c</td>
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<td>0.002</td>
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</tr>
</tbody>
</table>

a Differences among means were evaluated using the Tukey’s HSD (homoscedastic distribution) or the Tamhane’s T2 (heteroscedastic distribution) multiple comparison tests.

Table 2
Color parameters and pH values throughout cheese maturation. The results are presented as mean ± SD. Statistical classification was performed in each lot individually.a

<table>
<thead>
<tr>
<th>Lot</th>
<th>Maturation</th>
<th>L*</th>
<th>a*</th>
<th>b*</th>
<th>pH</th>
</tr>
</thead>
<tbody>
<tr>
<td></td>
<td></td>
<td>interior</td>
<td>interior</td>
<td>interior</td>
<td>interior</td>
</tr>
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<td>3.032.</td>
<td>Medium</td>
<td>85 ± 4 a</td>
<td>69 ± 3 b</td>
<td>−5 ± 1</td>
<td>18 ± 1 c</td>
</tr>
<tr>
<td></td>
<td>Long</td>
<td>82 ± 4 b</td>
<td>72 ± 3 a</td>
<td>−5 ± 1</td>
<td>19 ± 1 a</td>
</tr>
<tr>
<td></td>
<td>Extra-long</td>
<td>67 ± 1 c</td>
<td>72 ± 1 a</td>
<td>−5 ± 1</td>
<td>17 ± 1 b</td>
</tr>
<tr>
<td>Homoscedasticitya</td>
<td>p-value (n = 27)</td>
<td>&lt;0.001</td>
<td>&lt;0.001</td>
<td>0.005</td>
<td>&lt;0.001</td>
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<tr>
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<td>&lt;0.001</td>
<td>0.064</td>
<td>&lt;0.001</td>
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<tr>
<td>9.101.</td>
<td>Medium</td>
<td>69 ± 3 a</td>
<td>70 ± 1</td>
<td>−7 ± 1 c</td>
<td>3 ± 1 c</td>
</tr>
<tr>
<td></td>
<td>Long</td>
<td>65 ± 2 b</td>
<td>69 ± 2</td>
<td>−6 ± 1 b</td>
<td>−2 ± 0.5 b</td>
</tr>
<tr>
<td></td>
<td>Extra-long</td>
<td>68 ± 2 a</td>
<td>70 ± 1</td>
<td>−4 ± 0.2 a</td>
<td>−0.8 ± 0.2 a</td>
</tr>
<tr>
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<td>&lt;0.001</td>
<td>0.013</td>
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<tr>
<td>6.053.</td>
<td>Medium</td>
<td>67 ± 2 c</td>
<td>68 ± 1 b</td>
<td>−4 ± 0.4 b</td>
<td>−1 ± 1 c</td>
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<tr>
<td></td>
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<td>69 ± 2 b</td>
<td>68 ± 2 b</td>
<td>−5 ± 1 b</td>
<td>0 ± 1 b</td>
</tr>
<tr>
<td></td>
<td>Extra-long</td>
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<td>70 ± 2 a</td>
<td>−4 ± 1 a</td>
<td>1 ± 1 a</td>
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<tr>
<td>Homoscedasticitya</td>
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<td>One-way ANOVAa</td>
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<td>&lt;0.001</td>
<td>&lt;0.001</td>
<td>&lt;0.001</td>
<td>&lt;0.001</td>
</tr>
</tbody>
</table>

a Differences among means were evaluated using the Tukey’s HSD (homoscedastic distribution) or the Tamhane’s T2 (heteroscedastic distribution) multiple comparison tests.

b Homoscedasticity among formulations was tested by means of the Levene test: homoscedasticity, p-value>0.05; heteroscedasticity, p-value<0.05.

c When p<0.05 the mean value of the evaluated parameter of at least one maturation period is statistically different allowing performing multiple comparison tests. In each column, different letters indicate significantly different values among ripening periods within each lot.

As commonly observed among products of animal origin, saturated fatty acids (SFA) predominated (≥70%) over monounsaturated (MUFA) and polyunsaturated fatty acids (PUFA) (Table 3). The most abundant SFA were palmitic acid (21–27%), stearic acid (12–18%) and capric acid (8–12%), which were detected in higher percentages in cheese submitted to long and extra-long ripening. MUFA, on the other hand, and mostly due to oleic acid (17–24%), decreased along ripening, in line with the observed (except in lot 6.053) for linoleic and linolenic acids, the prevalent PUFAs.

As it could be expected, considering the extended ripening, fatty acids presented several statistically significant differences among the three assayed periods. Owing to the higher stability of SFA during storage, higher percentages after 18 and 24 months are reasonable. This does not necessarily mean that the absolute values of SFA increase; actually, the decrease in MUFA (especially C18:1) certainly contributed to the general increase in SFA, since the
Table 3

<table>
<thead>
<tr>
<th>Lot</th>
<th>Extra-long 4.2</th>
<th>Medium 3.9</th>
<th>Long 3.0</th>
<th>Extra-long 3.9</th>
<th>Medium 3.6</th>
<th>Long 2.7</th>
<th>Extra-long 2.4</th>
<th>Medium 2.1</th>
<th>Long 1.8</th>
</tr>
</thead>
<tbody>
<tr>
<td>6.053</td>
<td>0.5 a</td>
<td>0.001</td>
<td>0.1 b</td>
<td>0.1 b</td>
<td>0.001</td>
<td>0.1 b</td>
<td>0.001</td>
<td>0.1 b</td>
<td>0.1 b</td>
</tr>
<tr>
<td>9.01</td>
<td>0.5 a</td>
<td>0.001</td>
<td>0.1 b</td>
<td>0.1 b</td>
<td>0.001</td>
<td>0.1 b</td>
<td>0.001</td>
<td>0.1 b</td>
<td>0.1 b</td>
</tr>
</tbody>
</table>

3.2. Linear discriminant analysis

Results indicated in Tables 1–3 showed statistical significant differences among cheese submitted to different ripening periods for several parameters. Nevertheless, it would be useful to verify which differences characterized mostly each ripening period, as well as verifying if differences remained significant when evaluated simultaneously. With that purpose in mind, all results were compared together by LDA. The significant independent (p < 0.05) variables were selected by the stepwise procedure, considering Wilks’ λ test values.

The defined discriminant functions (Fig. 1) included 100.0% of the observed variance (first: 77.7%, second: 22.3%). These functions were defined according to the variables selected as mostly contributing to discriminate ripening periods. Among the 48 included variables, 19 were selected: saltiness, saponifiable matter, α° (all three in cheese interior and rind), moisture, protein, energy, C4:0, C10:0, C14:0, C18:0, C20:3 + C20:1, C22:0, C23:0 and C24:0 (all measured in cheese interior), as those having significant discriminant ability. Concerning correlations between selected variables and canonical discriminant functions, function 1 was mostly correlated with C20:3 + C21:0 (present only in medium-ripe samples), protein, saponifiable matter (cheese interior), C22:0 (all three lower in medium-ripe samples) and energy (higher in medium-ripe samples), separating samples corresponding to medium-ripe cheeses from those stored during longer periods (Fig. 1, horizontal axis). Function 2, on the other hand, was especially correlated with C23:0 (absent in extra-long ripened cheese), salt content (higher in long ripened cheese), saponifiable matter in rind and C22:6 (both higher in extra-long ripened cheese), separating markers corresponding to cheese samples submitted to long or extra-long ripening periods (Fig. 1, vertical axis). According to the leave-one-out classification, all samples were correctly classified, either for the originally grouped cases, as well as for cross-validated ones.

The high number of fatty acids among variables selected as having discriminant ability raised up the possibility that fatty acid profile alone could represent a chemical fingerprint to discriminate goat cheese according to ripening period. Therefore, a second linear discriminant analysis was tested including only the results obtained for fatty acids profiles. The two defined discriminant functions (Fig. 2) included also 100.0% of the observed variance (first: 81.8%, second: 18.2%). From the 26 individual fatty acids (plus SFA, MUFA and PUFA), only 13 were selected as having discriminant ability (C4:0, C10:0, C14:0, C16:0, C17:0, C18:0, C20:3, C20:1, C20:3 + C21:0, C20:5, C22:0, C23:0, C22:6 and C24:0). It is obvious that function separated primarily long and extra-long ripened cheese from medium ripe one, being mainly correlated to C4:0, C10:0 (lower in medium ripe cheese), C20:3 + C21:0 (only present in medium ripe cheese), C20:5 and C24:0 (both higher in medium ripe cheese). Regarding long and extra-long ripened cheese, which showed similar values for the fatty acids correlated with function 1, the corresponding markers were clearly separated by function 2, which was mostly correlated with C23:0 (absent in extra-long ripened cheese), C22:0 (higher in extra-long ripened cheese) and C22:6 (lower in extra-long ripened cheese). The classification performance was completely accurate (all samples were correctly classified, either for the originally grouped cases, as well as for cross-validated ones).
Fig. 1. Canonical discriminant functions coefficients defined from the evaluated parameters and plotted to highlight differences among ripening periods.

Fig. 2. Canonical discriminant functions coefficients defined according to fatty acids profiles and plotted to highlight differences among ripening periods.
4. Conclusion

Considering the evaluated parameters, several significant differences were found among cheese submitted to different ripening periods. In general, the assayed extended periods did not compromise the nutritional value of this highly appreciated product. Even so, the chemical changes that characterized mostly each ripening period (e.g., lower protein content, lower saponifiable matter and higher energy value in medium ripe cheese; higher salt percentage in long ripe cheese or higher saponifiable matter in extra-long ripe cheese) were effectively identified. Furthermore, fatty acids profile were attained as a potential chemical marker of cheese ripening duration, offering the possibility of discriminating different ripening periods in goat milk cheese by performing a single and simple chromatographic analysis.

Acknowledgments

The authors are grateful to the Foundation for Science and Technology (FCT, Portugal) and FEDER under Program PT2020 for financial support to CIMO (UID/AGR/00690/2013) and A. Fernandes (SFRH/BPD/114753/2016) grant. The authors are also grateful to the project funded by the Proder program - Measure 4.1 - Cooperation for Innovation - PA 49 481 - in partnership with LEICRAS.

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